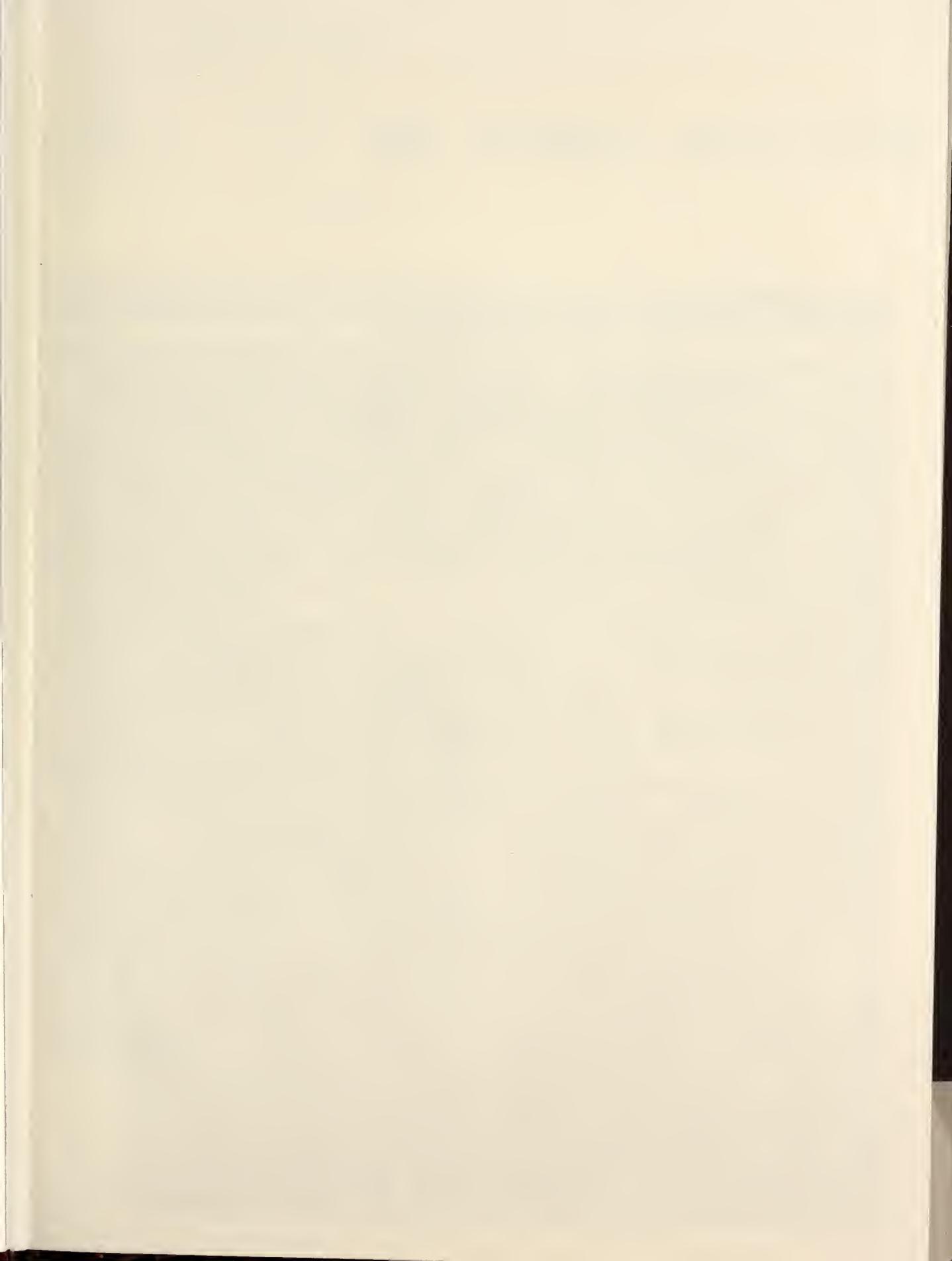


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Four Mathematical Models for the Prediction of LNG Densities

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Four Mathematical Models for the Prediction of LNG Densities

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FOUR MATHEMATICAL MODELS FOR THE PREDICTION OF LNG DENSITIES*

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Four mathematical models of the equation of state for LNG like mixtures are presented. The four models include an extended corresponding states model, a cell model, a hard sphere model and a revised Klosek and McKinley model. Each of the models has been optimized to the same experimental data set which included data for pure nitrogen, methane, ethane, propane, iso and normal butane, iso and normal pentane and mixtures thereof. For LNG like mixtures (mixtures of the orthobaric liquid state at temperatures of 120 K or less and containing at least 60% methane, less than 4% nitrogen, less than 4% each of iso and normal butane and less than 2% total of iso and normal pentane), all of the models are estimated to predict densities to within 0.1% of the true value. The revised Klosek and McKinley model is valid only for mixtures within the range of temperature and composition specified above while the other three models are valid for a broader range of pressure, temperature and composition. The experimental PVTx data set used in the optimization together with comparisons are given and listings of computer programs for each of the models are included.

Key words: Cell model; comparisons; computer programs; corresponding states; equation of state; hard sphere; LNG; mixtures; PVTx data; revised Klosek and McKinley.

1. INTRODUCTION

The purpose of this report is to present in final form the four mathematical models which were optimized to the experimentally determined orthobaric liquid

* This work was carried out at the National Bureau of Standards under the sponsorship of British Gas Corp., Chicago Bridge and Iron Co., Columbia Gas Service Corp., Distrigas Corp., Easco Gas LNG, Inc., El Paso Natural Gas, Gaz de France, Marathon Oil Co., Mobil Oil Corp., Natural Gas Pipeline Co., Phillips Petroleum Co., Shell International Gas, Ltd., Sonatrach, Southern California Gas Co., Tennessee Gas Pipeline, Texas Eastern Transmission Co., Tokyo Gas Co., Ltd., and Transcontinental Gas Pipe Line Corp., through a grant administered by the American Gas Association, Inc.

PVTx data of Miller and Hiza [25], Haynes and Hiza [12], Haynes, et al. [11], Hiza, et al. [14], Hiza and Haynes [15] and Haynes [9]. Interim results of the project have been reported by Haynes, et al. [13] and by McCarty [23]. The models reported in these two interim publications differ only slightly from those presented here. A companion archival document, McCarty [24] with the same results but in much less detail has been submitted to the Journal of Chemical Thermodynamics. The intent of the documentation here is more in the vein of a user's handbook. The above experimental data are for the liquid phase of nitrogen, methane, ethane, propane, normal and isobutane and various mixtures thereof.

The goal of the project (the project included the above referenced experimental work) was to produce one or more computer models which would predict the density of LNG to within 0.1% of the true value from a knowledge of the temperature, pressure and composition of the LNG. At the beginning of this study LNG was defined as mixtures of the above components (N_2 , CH_4 , C_2H_6 , C_3H_8 , n and iC_4H_{10}) and only the saturated liquid between 95 - 150 K was to be considered. Near the end of the project n and iC_5H_{12} were added to the list of allowable components but no experimental PVTx of pure n and iC_5H_{12} or binary systems containing n and iC_5H_{12} were measured as part of the project. The inclusion of these two components is based on data from Orrit, et al. [29] and Orrit, et al. [30].

Four models were considered: the extended corresponding states model, a hard sphere model, a cell model and a Revised Klosek and McKinley model. With the exception of the Revised Klosek and McKinley model only pure fluid and binary system data were used to optimize the models. In the case of the Revised Klosek and McKinley model multicomponent PVTx data were used in the optimization process.

Over a normal range of LNG composition and temperature all four of the models predict densities which agree to within 0.1% of experiment. This is true of all of the experimental PVTx measurements on LNG like mixtures made as part of this project.

No equation or mathematical model based on experimental data can be more accurate than the original data and therefore the extent to which the original goal of 0.1% accuracy has been met depends entirely upon the accuracy of the experimental data referenced above.

There is no reason to doubt the experimental data and therefore there is every reason to believe that the goal of the project has been achieved.

2. EXTENDED CORRESPONDING STATES

The thermodynamic equations for the extended corresponding states method are developed in a paper by Rowlinson and Watson [35] and only a very brief description will be given here. Leach [20] developed transformation functions for hydrocarbons which are called shape factor functions. Mollerup [27] and Mollerup and Rowlinson [26] combined the earlier work with the equation of state for methane by Goodwin [8] to produce a computer program to calculate the density of LNG mixtures, which was further modified by Mollerup [28].

The computer program in Appendix F for the calculation of LNG densities based on the extended corresponding states method is an extensive revision of the Mollerup program. Earlier revisions were reported by McCarty [23] and Haynes, Hiza and McCarty [13].

The extended corresponding states method is defined by the following equations:

$$z_i[P, T] = z_0[P \cdot h_{ii,o}/f_{ii,o}, T/f_{ii,o}] \quad (1)$$

$$G_i[P, T] = f_{ii,o} G_o[P, h_{ii,o}/f_{ii,o}, T/f_{ii,o}] - RT \ln(h_{ii,o}) \quad (2)$$

where Z is the compressibility factor, G is the Gibbs free energy, P is pressure and T is temperature. The subscript o denotes the reference fluid and the subscript i denotes the fluid for which properties are to be obtained via the equation of state for the reference fluid and the transformation functions $f_{ii,o}$ and $h_{ii,o}$. The double subscript ii is introduced now to allow extension of the method to mixtures. The two defining eqs (1) and (2) are necessary since there are two transformation functions. In this case the equation of state for methane by McCarty [22] was chosen for the reference fluid. During the course of the study it was necessary to modify the equation of state by McCarty [22] to give a realistic vapor liquid phase boundary down to a temperature of 43 K. This modification was necessary to accommodate the very low reduced temperatures of the heavier hydrocarbons and was accomplished without changing the performance of the equation of state above the triple point of methane. The equation of state is given in Appendix B.

The $f_{ii,o}$ and $h_{ii,o}$ are defined as

$$f_{ii,o} = (T_{ii}^c/T_o^c) \theta_{ii,o} (T_r, V_r) \quad (3)$$

and

$$h_{ii,o} = (V_{ii,o}^c/V_o^c) \phi_{ii,o} (T_r, V_r) \quad (4)$$

where

$$\theta_{ii,o} = 1 + (w_i - w_o)[n_1 - n_2 \ln T_{r_i} + (n_3 - n_4/T_{r_i})(V_{r_i} - n_5)] \quad (5)$$

and

$$\phi_{ii,o} = \frac{z_o^c}{z_i^c} \left[1 + (w_i - w_o) [n_6(v_{r_i} - n_7) - n_8(v_{r_i} - n_9) \ln T_{r_i}] \right] \quad (6)$$

The v_{r_i} and T_{r_i} are reduced temperature and volume, (i.e., $T_{r_i} = T/T_{ii}^c$ and

$v_{r_i} = v/v_{ii}^c$) each fluid requires a unique w_i which was estimated using pure fluid

experimental data. A single set of the n_i 's are used for all fluids. The n_i 's were estimated using all of the pure fluid experimental data from this study. The

z_o^c/z_i^c is the ratio of the compressibility factors ($z^c = P_c V_c / RT_c$) at the critical point. The parameters n_j , n_j , w_i and Z_{ii} are given in Appendix B. All of these parameters were estimated using the experimental PVT data set from this laboratory and least squares estimation techniques.

The extension of the above to mixtures is now accomplished by the following application of the following combining rules:

$$h_{x,o} = \sum_i \sum_j x_i x_j h_{ij,o} \quad (7)$$

$$f_{x,o} h_{x,o} = \sum_i \sum_j x_i x_j f_{ij,o} h_{ij,o} \quad (8)$$

$$f_{ij,o} = \xi_{ij} (f_{ii,o} f_{jj,o})^{1/2} \quad (9)$$

$$h_{ij,o} = n_{ij} (\frac{1}{2} h_{ii,o}^{1/3} + \frac{1}{2} h_{jj,o}^{1/3})^3 \quad (10)$$

The ξ_{ij} and the n_{ij} are binary interaction parameters determined by least squares from the PVTx data for binary mixtures. These parameters are given in Appendix B.

This method works quite well as may be seen in the comparisons in Appendix A. It has indeed reproduced all of the present experimental data set to within $\pm 0.1\%$ except for 14 out of a total of 285 experimental data points. Of these 14 points, 11 are judged to have an uncertainty greater than 0.1%. Figure 1 presents the deviations between the calculated and experimental densities for these 14 points. Appendix A contains comparisons of calculated and experimental densities for the entire data set. This is the best performance of the four models presented here. No pressure, temperature or composition restrictions have been placed on this model.

In the interim publications by McCarty [23] and Haynes, Hiza and McCarty [13] some doubt about the accuracy of the calculated densities was expressed because of the disagreement with a few binary and multicomponent systems containing methane and butane. This disagreement has since been resolved by additional measurements (Haynes [9], Haynes [10] and Miller and Hiza [25] on some of the systems which agree with the predictions of the model but disagree with the previous measurements. The net result of the new measurements is a very slight change in binary interaction coefficients of the methane-butane and nitrogen-butane system. These changes have no practical effect on LNG like mixtures where the concentrations of N_2 , iC_4H_{10} and nC_4H_{10} are individually less than 5%. In other words either the models presented here or those in the interim publications may be used to predict the density of a LNG like mixture to within 0.1% of the true density.

3. A HARD SPHERE METHOD

The model of Rodosevich and Miller [33] is one of many modifications of the Longuet-Higgins and Widom [21] model, and was chosen to be included in this study

as a representative example of the application of the hard sphere equation of state concept to the correlation of PVTx data. The equation of state by Rodosevich and Miller [33] is

$$\frac{PV}{RT} = c \frac{1 + y + y^2}{(1 - y)^3} - \frac{a}{RTV} \quad (11)$$

where the $y = b/4V$ and a , b , and c are adjustable parameters, P is pressure, V is specific volume, T is temperature and R is the gas constant. The equation is applied to mixtures by assuming the one-fluid theory and applying the following combining rules.

$$a_m = \sum_i \sum_j a_{ij} x_i x_j \quad (12)$$

$$b_m = \sum_i \sum_j b_{ij} x_i x_j \quad (13)$$

$$c_m = \sum_i \sum_j c_{ij} x_i x_j \quad (14)$$

The mixing rules are:

$$b_{ij} = \left[\frac{b_{ii}^{1/3} + b_{jj}^{1/3}}{2} (1 - j_{ij}) \right]^3 \quad (15)$$

$$a_{ij} = (a_{ii} a_{jj})^{1/2} \left[\frac{b_{ij}^2}{b_{ii} b_{jj}} \right]^{1/2} (1 - k_{ij}) \quad (16)$$

$$c_{ij} = \frac{c_{ii} + c_{jj}}{2} \quad (17)$$

The parameters j_{ij} and k_{ij} are in this case the binary interaction parameters. The a 's, b 's, c 's, j_{ij} 's and k_{ij} 's are given in Appendix B. The excess volume is now calculated using the equation of state and

$$V_E = \tilde{V}_m - \tilde{V}_i x_i \quad (18)$$

where \tilde{V}_m and the \tilde{V}_i are calculated via the eqs 11 through 17 and then

$$V_m = V_i x_i + V_E \quad (19)$$

where the V_E is from eq (18) and the V_i are from experimental data. The values of V_i in this case were calculated from the equations for the liquid density of the pure fluids given in Appendix C.

The above equations are those of Rodosevich and Miller [33] and Rodosevich [34] and only the j_{ij} 's and k_{ij} 's have been revised on the basis of the present new data set, and only binary systems data were used to estimate via least squares the j_{ij} 's and k_{ij} 's.

As the method is used here it is an excess volume method, and consequently when the temperature of the mixture approaches the critical temperature of one of the component fluids, the method fails. Since the critical temperature of nitrogen is about 126 K, this method should not be used for mixtures containing nitrogen at temperatures above 120 K. Eliminating the data points for mixtures which contain nitrogen at temperatures above 120 K reduces the set from 285 to 251 PVTx points. Figure 2 is a percentage deviation plot containing all of the data points from the set of 251 for which densities calculated by the hard sphere method differ from the experimental density by more than 0.1%. Two things are readily seen in comparing figs. 1 and 2; first, even though total number of points has been reduced in the comparison set, the number of points for which deviations exceed 0.1% in the hard sphere comparison, fig. 2, is far more than for the extended corresponding states comparison, fig. 1. Second, the hard sphere method becomes more uncertain for all mixtures, regardless of components as the temperature exceeds 115 K.

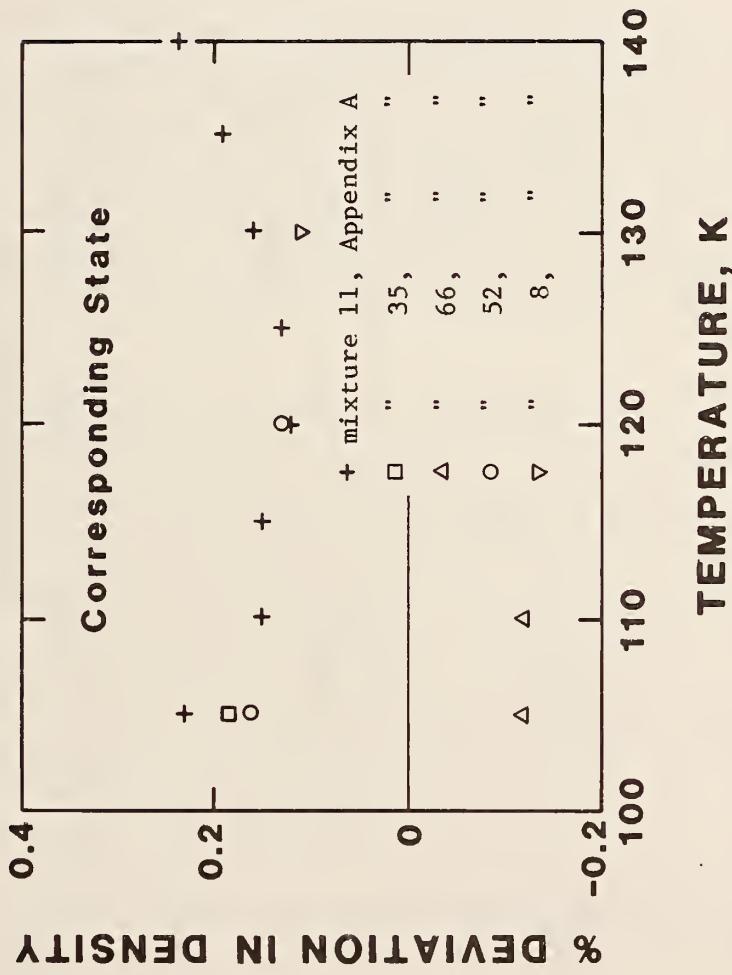
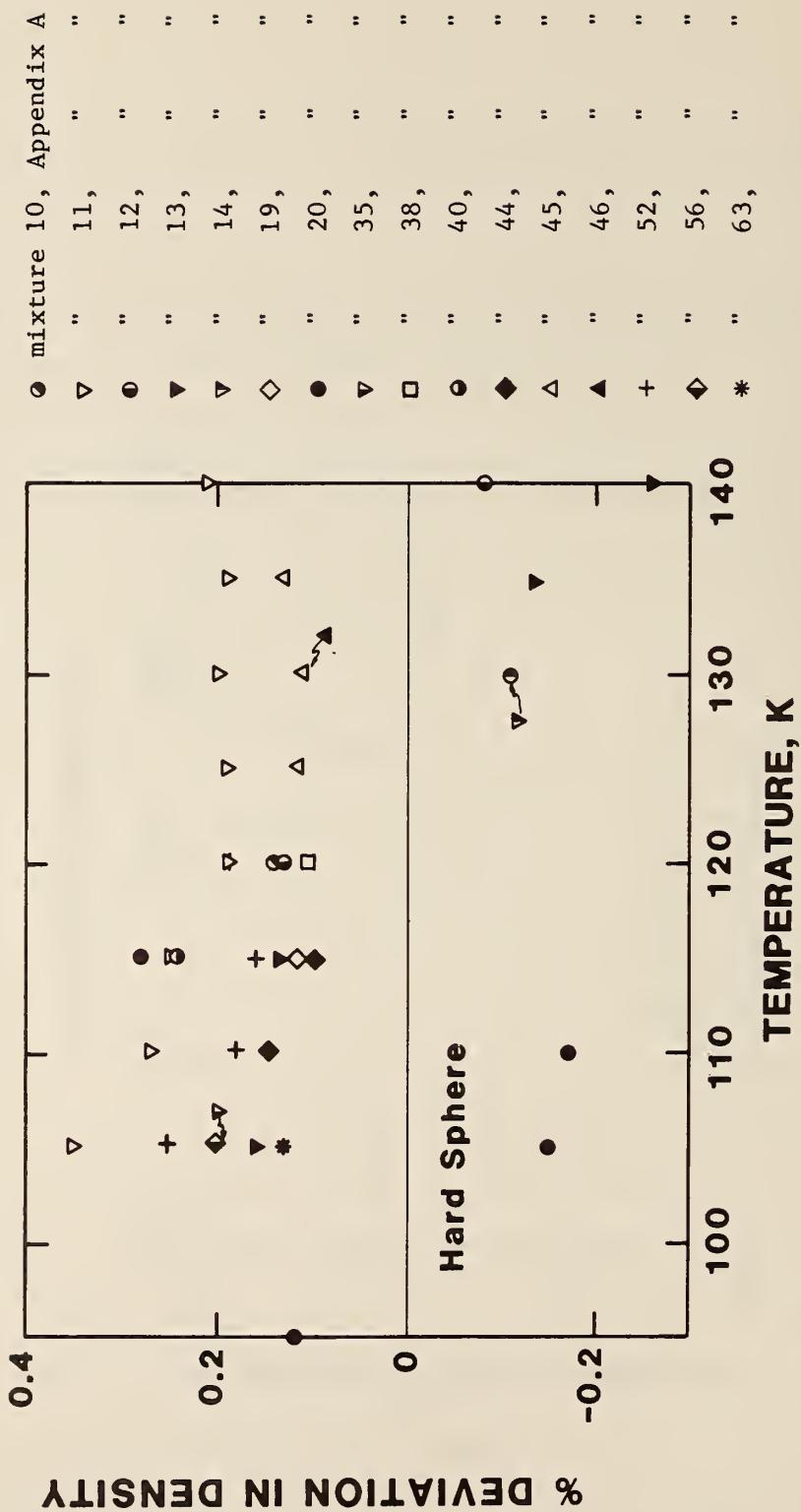


Figure 1. All deviations greater than 0.1% between experimental and calculated densities using the Extended Corresponding States method. The comparison set is all of the data points in Appendix A. (285 data points)



Deviations greater than 0.1% between experimental and calculated densities using the Hard Sphere model. The comparison set is all of the data in Appendix A except those data points for mixtures containing N₂ at temperatures above and including 120 K. (251 data points)

4. A REVISED KLOSEK AND MCKINLEY METHOD

The Klosek and McKinley method [18] is a totally empirical recipe for calculating the density of a LNG-like mixture given the temperature and composition. Pressure is not taken into account. However, this does not seem to be a serious omission. The procedure proposed by Klosek and McKinley [18] is as follows:

$$V_{\text{mix}} = \sum X_i V_i - k X_{\text{CH}_4} \quad (20)$$

where V_{mix} is the volume of the mixture, X_i and V_i are the mole fraction and volume of the i^{th} component, X_{CH_4} is the mole fraction of methane and k is a correction factor obtained from a table or graph. The V_i and k are obviously temperature dependent and in addition k is dependent upon the molecular weight of the mixture.

Using the present data set k was calculated for all of the experimental data points where methane was present in the mixture and excluding all data points where N_2 was present in greater than 5% concentration. Figure 3 shows a typical isotherm for k , with N_2 present (labeled k_2) and without N_2 present (labeled k_1). All of the isotherms available show similar behavior, i.e., all of the k 's for mixtures containing nitrogen (of about 5%) fall on one line and all of those for mixtures without nitrogen fall on another. Since all of the mixtures with nitrogen have about the same amount of nitrogen present (about 4.5%), the method was modified by adding a term to take into account the nitrogen when it is present. The equation becomes

$$V_{\text{mix}} = \sum X_i V_i - [k_1 + (k_2 - k_1) X_{N_2} / .0425] X_{\text{CH}_4} \quad (21)$$

where everything is the same as in eq (20) except that k_1 is read from one

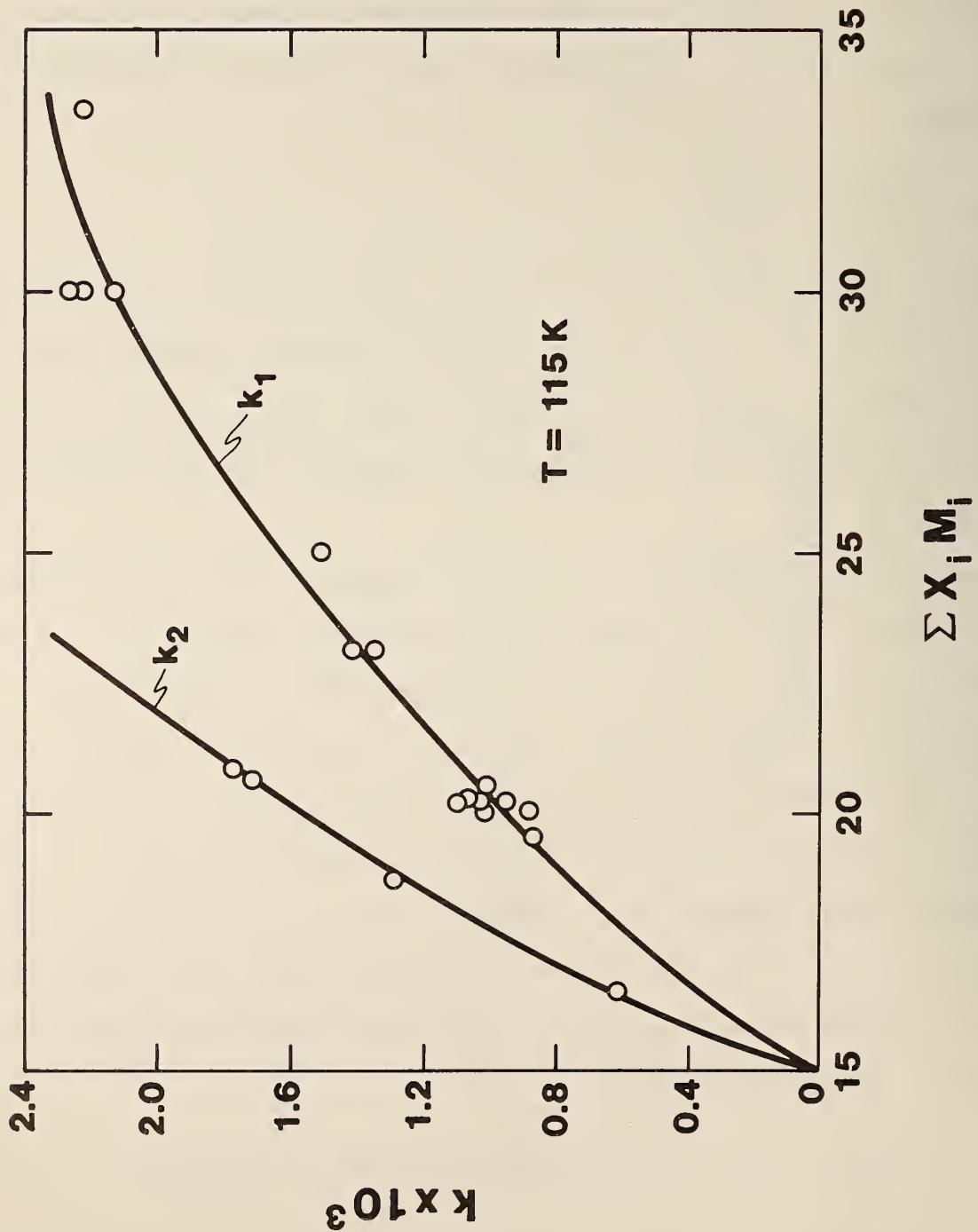


Figure 3. A plot of the correction factors (k_1 and k_2) for the 115 kelvin isotherms.

curve and k_2 is read from the other. Appendix D gives tables of values for the V_i , k_1 and k_2 which are spaced such that linear interpolation is adequate in both variables (i.e., temperature or molecular weight). The k factors in Appendix D have been obtained graphically from the multicomponent PVTx data of Hiza and Haynes [15] and Miller and Hiza [25] as well as densities calculated from the extended corresponding states method of section 2.

The limits of compositions of the revised Klosek and McKinley method are the most severe of any of the methods given here. This method should not be used for mixtures other than LNG like mixtures and for LNG like mixtures only when they contain at least 60% methane, less than 4% nitrogen, less than 4% each of iC_4H_{10} , nC_4H_{10} and less than 2% total of iC_5H_{12} and nC_5H_{12} .

There are 40 experimental PVTx points from the original set of 285 which may be considered LNG like and fall within the composition limits outlined above. Figure 4 shows all of the deviations between calculated and experimental densities in this 40 point comparison set which exceeds the 0.1% criterion. The deviation trends for the revised Klosek and McKinley method (fig. 4) are very similar to those of the hard sphere method (fig. 2) and in fact all of the deviations in fig. 4 occur at temperatures at or above 115 K, therefore the method can only be considered as accurate as the others for LNG like mixtures at temperatures below 115 K.

5. THE CELL MODEL

The cell model considered here was originally proposed by Renon, et al. [32]. In a paper by the same three authors which appeared simultaneously (Eckert, et al. [7]), the cell model was applied to mixtures via Scott's [36] two-fluid theory and a three parameter corresponding states theory. Albright [2] further modified the method by modifying the mixing rules on the basis of a

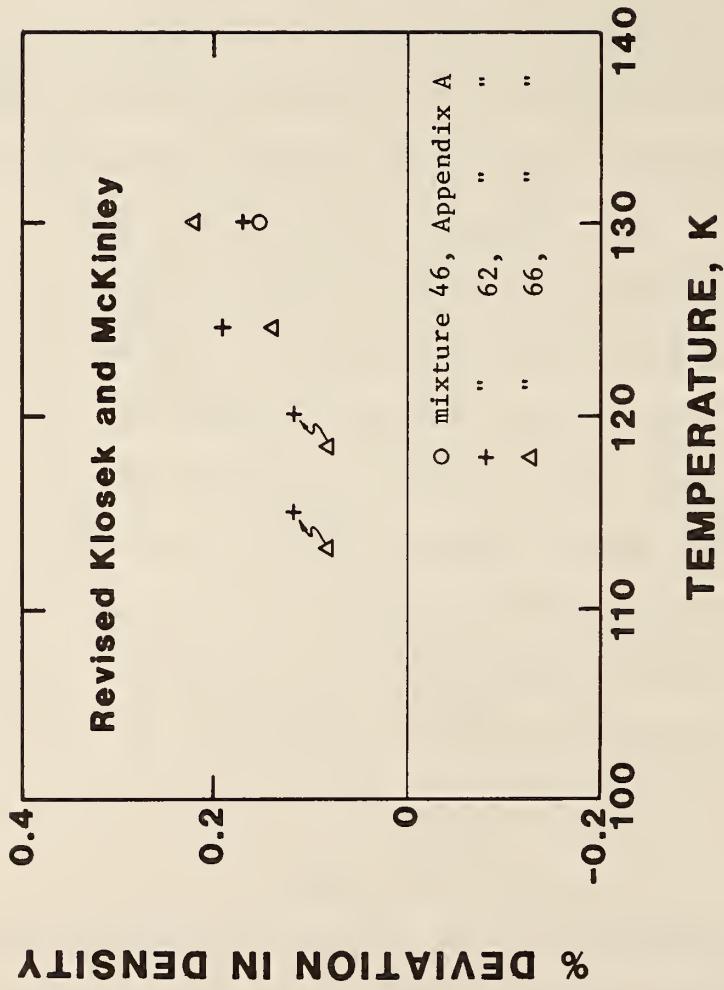


Figure 4. All deviations greater than 0.1% between experimental and calculated densities using the Revised Klosek and McKinley model. The comparison set is all multicomponent mixture data in Appendix A with $> 60\%$ CH_4 , $< 4\%$ nC_4H_{10} , $< 4\%$ iC_4H_{10} , $< 4\%$ N_2 and $< 2\%$ of $\text{n} + \text{iC}_5\text{H}_{12}$. (40 data points)

proposal by Yuan [38] and by inserting a pressure dependence based on the experimental liquid ethane data by Pope [31].

The optimization of this method was carried out by M. Albright [1] at Phillips Petroleum Company in Bartlesville, Oklahoma and the details of this work will be published elsewhere. The model is included here because it was optimized to the same data set as the others and therefore the comparisons between experimental and calculated densities given here in fig. 5 together with figs. 1, 2 and 4 provide a common basis of comparison with the other three methods. A listing of the computer program is given in Appendix F.

The same data set as was used in the hard sphere method for comparison has been used here, i.e., all of the data points for mixtures containing nitrogen at temperatures 120 K and above have been taken out of the original 285 points leaving a total of 251 data points.

As in the case of the other methods fig. 5 shows all of the points for which the calculated and experimental densities differ by more than 0.1%.

6. USE OF THE METHODS

When the project started in 1972, the atomic weights of nitrogen, carbon and hydrogen were taken from the 1961 carbon 12 scale, IUPAC [16]. During the course of the investigation a revision, Atomic Weights of the Elements [3], to this scale appeared. The revision changed slightly the atomic weights of carbon and hydrogen, but since the changes were small (the maximum difference in any of the densities used here is 0.003%), and because changing the atomic weights would not change the relative results, the changes were not made. Therefore when using the tables and programs in the appendices, the molecular weights given in the tables and programs should be used to maintain consistency.

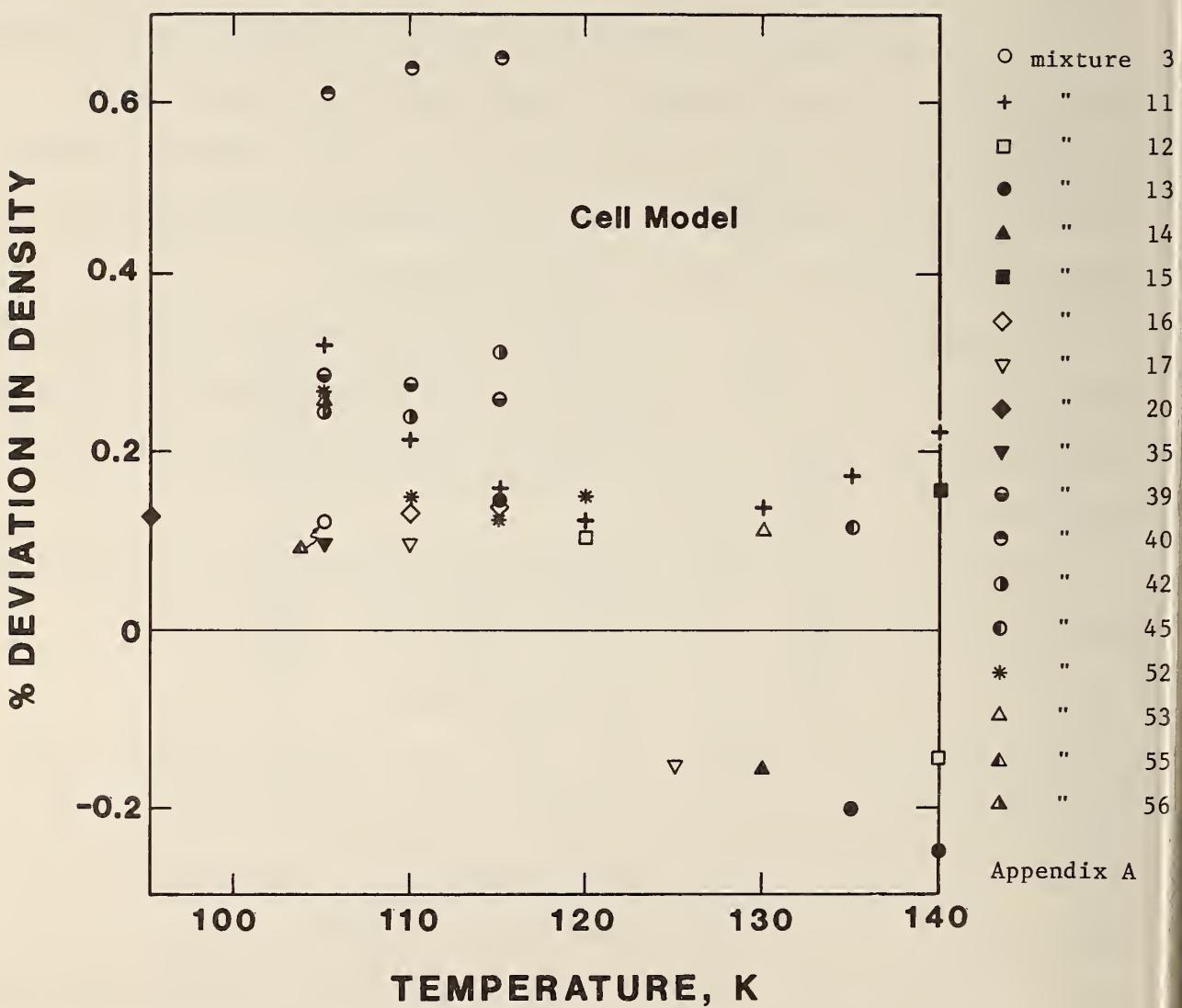


Figure 5. Deviations greater than 0.1% between experimental and calculated densities using the Cell model. The comparison set is all of the data in Appendix A except those data points for mixtures containing N₂ at temperatures above and including 120 K. (251 data points)

The critical parameters used here are from: CH₄, McCarty [22]; C₂H₆, Sliwinski [37]; C₃H₈, Das, et al. [4]; iC₄H₁₀, Das, et al. [5]; nC₄H₁₀, Das, et al. [6]; iC₅H₁₂, Kudchadker, et al. [19]; and N₂, Jacobsen, et al. [17].

Errors in the input variables will of course, cause errors in the density predicted by the models. In general, the error in density caused by an error in the input variables is a function of those input variables, and must be treated on an individual basis. However, for LNG like mixtures certain general trends are found. An error in the pressure must be at least 50% before it will have any effect at all on the resulting density. An error in composition, unless it is of the order of several percent, will cause the same relative error in density as it will cause in the molecular weight of the mixture, i.e., if an error in composition causes a 0.1% error in the resulting molecular weight, it will also cause a 0.1% error in the predicted density.

The error in the calculated density due to an error in the input temperature is a function of the composition and the temperature. Table 1 gives resulting errors in density for a 1% error in temperature, for three hypothetical LNG like mixtures.

In general the errors in density caused by an error in temperature are the largest for mixtures containing a high concentration of the most volatile fluids, CH₄ and N₂, and correspondingly the errors decrease as the concentration of the heavier hydrocarbons increases in the mixture. These errors are not a function of which model is being used.

When using the extended corresponding states method, one should keep in mind that twelve significant figures are required by the methane equation of state. The hard sphere model also uses the methane equation from McCarty [22] and the nitrogen equation of Jacobsen, et al. [17] to calculate compressibilities and

TABLE 1. Errors in Density Caused by an Error in the
Input Temperature of 1%.

Temperature K	% Error in Density		
	Mix A [*]	Mix B [*]	Mix C [*]
95	0.28	0.25	0.20
100	0.30	0.27	0.22
105	0.32	0.29	0.24
110	0.35	0.32	0.29
115	0.39	0.34	0.31

Mix A^{*} = 0.95 CH₄, 0.05 N₂

Mix B^{*} = 0.9 CH₄, 0.02 C₂H₆, 0.02 C₃H₈, 0.02 iC₄H₁₀, 0.02 nC₄H₁₀, 0.02 N₂

Mix C^{*} = 0.6 CH₄, 0.3 C₂H₆, 0.02 C₃H₈, 0.02 iC₄H₁₀, 0.02 nC₄H₁₀, 0.02 N₂,
0.02 iC₅H₁₂.

*Arbitrary LNG like compositions assumed for the purpose of illustrating the effect of an error in the input temperature.

therefore requires twelve significant figures to insure the accuracy of the calculated density. The other two models require only eight significant figures to be carried along in the calculations.

7. CONCLUSIONS

On the basis of the performance of the four models given here and subject to the composition and temperature restrictions already noted, it is estimated that given the pressure, temperature and composition of LNG, any one of the four models may be used to predict the density to within 0.1% of the true value. As has already been mentioned (see section 1) the above accuracy statement is dependent entirely upon the accuracy of the experimental data in Haynes, et al. [11], Haynes, et al. [13], Hiza, et al. [14], Haynes [9], Hiza and Haynes [15], Miller and Hiza [25] and Haynes [10]. These data have been estimated by the authors to be accurate to within 0.1% of the true value with a precision of a few hundredths of a percent. The work on the models given here have provided no basis for questioning the claims of the experimenters, in fact the ability of the models to predict the densities of the multicomponent mixtures to within 0.1% of the measured values tends to support the accuracy claims of the experimenters.

Interim results of this study were reported by Haynes, et al. [13] and McCarty [23], both of which contain earlier versions of the mathematical models given here. These earlier versions are only slightly different than the final ones and for the purposes of calculating LNG densities either of the versions may be used. The reader is, however, cautioned to read the limitations of each model as defined in the earlier sections.

Computer programs for the four models are available at the Thermophysical Properties Division of the National Bureau of Standards in Boulder, Colorado.

8. REFERENCES

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Appendix A. Experimental Data

The following is a list of all of the experimental PVTx mixtures data which were measured during the course of this project. The data are from Miller and Hiza [25]; Haynes and Hiza [12]; Haynes, et al. [11]; Hiza, et al. [14]; Hiza and Haynes [15] and Haynes [9,10].

All of the data are for the orthobaric liquid except for the data of Miller and Hiza [25] which are for the single phase liquid phase, very close to the orthobaric conditions. The units of the data are bars, moles per liter and kelvin. The columns labeled RKM, HS, CELL and CS correspond to percentage derivations between experimental and predicted densities by the Revised Klosek and McKinley, hard sphere, cell and extended corresponding states models respectively. The derivations are always calculated using experimental-calculated densities.

MIXTURE NO 1 (Miller and Hiza [25])

0.85147 CH₄ + 0.14853 C₂H₆

P	D	T	MW	RKM	HS	CELL	CS
.977	25.7499	110.08	18.1265	.13	- .02	.02	.01

MIXTURE NO 2 (Hiza, et al. [14])

0.68006 CH₄ + 0.31994 C₂H₆

.416	25.1027	105.00	20.5309	.14	.05	.10	.07
.645	24.7802	110.00	20.5309	.07	- .01	.02	.01
.961	24.4612	115.00	20.5309	.05	- .04	-.03	-.03
1.380	24.1402	120.00	20.5309	.06	- .06	-.06	-.05
1.930	23.8212	125.00	20.5309	.08	- .05	-.04	-.05
2.620	23.5007	130.00	20.5309	.12	- .01	.00	-.02

MIXTURE NO 3 (Hiza, et al. [14])

0.49325 CH₄ + 0.50675 C₂H₆

.325	23.9619	105.00	23.1513	.14	.07	.11	.07
.503	23.6937	110.00	23.1513	.09	.03	.03	.03
.749	23.4328	115.00	23.1513	.11	.03	.01	.03
1.080	23.1559	120.00	23.1513	.07	- .02	-.06	-.03
1.500	22.8933	125.00	23.1513	.08	.00	-.05	.00
2.050	22.6290	130.00	23.1513	.14	.04	-.02	.03
2.720	22.3581	135.00	23.1513	.19	.06	.01	.05
3.550	22.0765	140.00	23.1513	.25	.05	.02	.04

MIXTURE NO 4 (Hiza, et al. [14])

0.35457 CH₄ + 0.64543 C₂H₆

.256	23.1033	105.00	25.0965	- .04	- .07	-.04	-.07
.397	22.8777	110.00	25.0965	- .01	- .04	-.05	-.05
.580	22.6478	115.00	25.0965	.03	- .02	-.06	-.03
.826	22.4035	120.00	25.0965	.00	- .06	-.11	-.07
1.146	22.1872	125.00	25.0965	.10	.04	-.03	.03
1.550	21.9441	130.00	25.0965	.10	.03	-.04	.02

MIXTURE NO 5 (Hiza, et al. [14])

0.85796 CH₄ + 0.14204 C₃H₈

P	D	T	MW	RKM	HS	CELL	CS
.517	24.9622	105.00	20.0279	.01	.10	.10	.07
.817	24.6332	110.00	20.0279	- .02	.08	.07	.06
1.199	24.2942	115.00	20.0279	- .04	.04	.03	.03
1.785	23.9492	120.00	20.0279	- .07	.00	.00	.00
2.415	23.5942	125.00	20.0279	- .09	- .06	- .04	- .05
3.290	23.2461	130.00	20.0279	- .11	- .06	- .01	- .04

MIXTURE NO 6 (Hiza, et al. [14])

0.74920 CH₄ + 0.25080 C₃H₈

.478	23.4767	105.00	23.0790	.01	.02	.03	-.02
.738	23.2064	110.00	23.0790	- .03	.00	-.01	-.03
1.099	22.9364	115.00	23.0790	- .01	.00	-.02	-.02
1.582	22.6665	120.00	23.0790	.02	.01	-.01	.01
2.216	22.3818	125.00	23.0790	- .03	- .04	-.04	-.01
3.029	22.1019	130.00	23.0790	.02	- .05	-.02	.01

MIXTURE NO 7 (Hiza, et al. [14])

0.49637 CH₄ + 0.50363 C₃H₈

.384	20.4909	105.00	30.1720	- .05	.01	.04	-.03
.591	20.3046	110.00	30.1720	- .08	.01	.01	-.02
.874	20.1180	115.00	30.1720	- .06	.01	-.02	-.02
1.250	19.9311	120.00	30.1720	.00	.01	-.03	.00
1.730	19.7471	125.00	30.1720	.04	.03	-.01	.04
2.320	19.5546	130.00	30.1720	- .01	.01	-.03	.05

MIXTURE NO 8 (Hiza, et al. [14])

0.29538 CH₄ + 0.70462 C₃H₈

.271	18.5132	105.00	35.8106	- .19	- .08	-.05	-.12
.409	18.3624	110.00	35.8106	- .24	- .09	-.10	-.12

MIXTURE NO 9 (Haynes [10])

0.92788 CH₄ + 0.07212 nC₄H₁₀

P	D	T	MW	RKM	HS	CELL	CS
1.820	24.2615	120.00	19.0779	.00	.04	-.04	-.04
2.547	23.8868	125.00	19.0779	.00	.00	-.07	-.05
3.470	23.5047	130.00	19.0779	-.01	-.03	-.09	-.07
4.616	23.1225	135.00	19.0779	.04	-.03	-.06	-.05
6.023	22.7284	140.00	19.0779	.16	-.04	-.03	-.03

MIXTURE NO 10 (Haynes [10])

0.92780 CH₄ + 0.07220 nC₄H₁₀

1.270	24.6285	115.00	19.0813	.01	.08	.01	-.01
1.824	24.2783	120.00	19.0813	.07	.11	.03	.04
2.549	23.8999	125.00	19.0813	.06	.07	-.01	.01

MIXTURE NO 11 (Hiza, et al. [14])

0.91674 CH₄ + 0.08326 nC₄H₁₀

.521	25.1536	105.00	19.5467	.21	.35	.33	.22
.810	24.7960	110.00	19.5467	.12	.26	.21	.15
1.216	24.4512	115.00	19.5467	.14	.24	.18	.15
1.753	24.0889	120.00	19.5467	.10	.19	.11	.11
2.472	23.7370	125.00	19.5467	.15	.18	.10	.13
3.374	23.3789	130.00	19.5467	.17	.19	.12	.16
4.509	23.0110	135.00	19.5467	.21	.19	.15	.18
5.887	22.6391	140.00	19.5467	.35	.20	.22	.23

MIXTURE NO 12 (Haynes [10])

0.77982 CH₄ + 0.22018 nC₄H₁₀

1.702	21.6066	120.00	25.3805	.12	.14	.11	.06
2.369	21.3549	125.00	25.3805	.11	.08	.02	.04
3.228	21.1020	130.00	25.3805	.09	.00	-.07	.03
4.291	20.8555	135.00	25.3805	.12	-.03	-.10	.07
5.576	20.5978	140.00	25.3805	.20	-.13	-.15	.07

MIXTURE NO 13 (Haynes [10])

 0.77762 CH₄ + 0.22238 nC₄H₁₀

P	D	L	T	MW	RKM	HS	CEL	CS
1.179	21.8054		115.00	25.4011	.07	.15	.16	.02
1.699	21.5601		120.00	25.4011	.07	.10	.07	.02
2.372	21.3164		125.00	25.4011	.10	.06	.01	.03
3.230	21.0605		130.00	25.4011	.06	-.03	-.10	.00
4.291	20.8011		135.00	25.4011	.02	-.14	-.20	-.03
5.579	20.5448		140.00	25.4011	.09	-.23	-.26	-.03

MIXTURE NO 14 (Hiza, et al. [14])

 0.58828 CH₄ + 0.41172 nC₄H₁₀

3.183	18.3058	130.00	33.3687	- .32	- .11	-.15	-.08
2.342	18.4853	125.00	33.3687	- .17	.00	.00	-.03
2.281	18.4772	125.00	33.3687	- .21	-.05	-.04	-.08
1.636	18.6495	120.00	33.3687	- .18	.02	.07	-.06

MIXTURE NO 15 (Haynes [10])

 0.92044 CH₄ + 0.07956 iC₄H₁₀

1.254	24.3633	115.00	19.3910	- .21	.05	.06	-.01
1.805	24.0029	120.00	19.3910	- .24	.02	.03	-.03
2.521	23.6403	125.00	19.3910	- .22	.01	.04	-.03
3.434	23.2752	130.00	19.3910	- .21	.03	.08	.00
4.567	22.8920	135.00	19.3910	- .22	.00	.09	-.02
5.950	22.5037	140.00	19.3910	- .14	-.02	.14	-.02

MIXTURE NO 16 (Haynes [10])

 0.78329 CH₄ + 0.21671 iC₄H₁₀

.782	21.9144	110.00	25.1625	- .33	.10	.15	.01
1.164	21.6652	115.00	25.1625	- .33	.05	.07	-.01
1.671	21.4136	120.00	25.1625	- .35	.01	.00	-.02
2.329	21.1668	125.00	25.1625	- .34	.00	-.03	.00
3.170	20.9125	130.00	25.1625	- .35	-.04	-.07	.00
4.208	20.6629	135.00	25.1625	- .33	-.06	-.06	.04
5.474	20.4082	140.00	25.1625	- .23	-.09	-.04	.07

MIXTURE NO 17 (Hiza, et al. [14])

0.48687 CH₄ + 0.51313 iC₄H₁₀

P	D	T	MW	RKM	HS	CELL	CS
.629	17.3575	110.00	37.6362	- .62	.05	.11	.02
.938	17.2076	115.00	37.6362	- .70	.00	-.01	-.02
1.361	17.0639	120.00	37.6362	- .64	-.02	-.08	-.02
1.852	16.9156	125.00	37.6362	- .69	-.07	-.17	-.04

MIXTURE NO 18 (Hiza, et al. [14])

0.95248 CH₄ + 0.04752 N₂

1.380	26.8476	105.00	16.6119	- .05	-.04	-.04	-.02
1.990	26.4052	110.00	16.6119	.06	.00	.00	.04
2.634	25.9374	115.00	16.6119	.20	.01	.02	.06
3.500	25.4522	120.00	16.6119	.26	*	.03	.07
4.600	24.9496	125.00	16.6119	.54	*	.04	.07
5.830	24.4210	130.00	16.6119	.69	-5.69	.03	.05
7.300	23.8600	135.00	16.6119	.77	-3.99	-.01	-.03
9.200	23.2809	140.00	16.6119	1.09	-2.69	-.02	-.07

MIXTURE NO 19 (Hiza, et al. [14])

0.69651 CH₄ + 0.30349 N₂

3.450	26.8735	100.00	19.6759	- .39	.06	.09	.00
4.661	26.3393	105.00	19.6759	- .56	.02	.07	.04
6.181	25.7686	110.00	19.6759	- .92	-.05	-.02	.00
8.010	25.1790	115.00	19.6759	-1.28	.13	-.08	-.03
10.112	24.5737	120.00	19.6759	-1.56	1.88	-.07	-.04

MIXTURE NO 20 (Hiza, et al. [14])

0.50758 CH₄ + 0.49242 N₂

3.303	27.0801	95.00	21.9375	-2.16	.11	.14	.06
4.468	26.4588	100.00	21.9375	-1.92	-.01	.05	.04
6.383	25.8106	105.00	21.9375	-1.56	-.15	-.05	-.01
8.461	25.1387	110.00	21.9375	-1.29	-.17	-.09	-.03
10.740	24.4431	115.00	21.9375	1.12	.28	-.04	-.01
13.983	23.7096	120.00	21.9375	-1.79	3.41	.07	-.01
17.529	22.9315	125.00	21.9375	-1.14	*	.26	.00
21.076	22.1005	130.00	21.9375	-.72	-2.84	.05	.10

* The hard sphere solution for the density of N₂ failed.

MIXTURE NO 21 (Hiza, et al. [14])

0.67287 C₂H₆ + 0.32713 C₃H₈

P	D	T	MW	RKM	HS	CELL	CS
	18.6192	125.00	34.6588	.00	.02	-.02	.00
	18.4648	130.00	34.6588	.00	.02	.00	.01
	18.3059	135.00	34.6588	.00	.00	.00	.00
	18.1509	140.00	34.6588	.00	.01	.04	.02

MIXTURE NO 22 (Hiza, et al. [14])

0.50105 C₂H₆ + 0.49895 C₃H₈

18.3618	105.00	37.0689	.02	.04	.02
18.2169	110.00	37.0689	.01	.00	.00
18.0726	115.00	37.0689	.00	-.03	-.01
17.9282	120.00	37.0689	-.01	-.05	-.01
17.7880	125.00	37.0689	.00	-.03	.00
17.6412	130.00	37.0689	-.02	-.04	-.02
17.4988	135.00	37.0689	-.01	-.01	-.01
17.3526	140.00	37.0689	-.03	.00	-.01

MIXTURE NO 23 (Hiza, et al. [14])

0.67117 C₂H₆ + 0.32883 nC₄H₁₀

17.5047	110.00	39.2952	.00	.02	.00
17.3706	115.00	39.2952	.01	.00	.00
17.1031	125.00	39.2952	.02	.00	.02
16.9626	130.00	39.2952	-.01	-.03	-.02
16.8285	135.00	39.2952	.00	-.01	-.01
16.6947	140.00	39.2952	.00	.03	.01

MIXTURE NO 24 (Hiza, et al. [14])

0.65343 C₂H₆ + 0.34657 nC₄H₁₀

17.2184	115.00	39.7929	.00	.00	.00
17.0824	120.00	39.7929	-.02	-.03	-.02

MIXTURE NO 25 (Hiza, et al. [14])

0.72436 C₂H₆ + 0.27564 iC₄H₁₀

17.9779	105.00	37.8030	.01	.01	.01
17.8401	110.00	37.8030	.01	.00	.01
17.4235	125.00	37.8030	.01	.00	.01
17.2825	130.00	37.8030	.00	.01	.01

MIXTURE NO 26 (Hiza, et al. [14])

0.68939 C₂H₆ + 0.31061 iC₄H₁₀

P	D	T	MW	RKM	HS	CELL	CS
	17.3716	115.00	38.7840		.00	-.03	.00
	17.2344	120.00	38.7840		-.02	-.04	-.02

MIXTURE NO 27 (Hiza, et al. [14])

0.94067 C₂H₆ + 0.05933 N₂

3.850	21.4718	105.00	29.9481	-.02	-.01	-.03
4.630	21.2912	110.00	29.9481	.03	.02	.03
5.472	21.0845	115.00	29.9481	-.01	-.04	-.04
6.383	20.8998	120.00	29.9481	-.72	.01	.00

MIXTURE NO 28 (Hiza, et al. [14])

0.60949 C₃H₈ + 0.39051 nC₄H₁₀

	14.6487	115.00	49.5749	-.01	-.02	-.01
	14.5521	120.00	49.5749	.02	.00	.02

MIXTURE NO 29 (Hiza, et al. [14])

0.60650 C₃H₈ + 0.39350 nC₄H₁₀

	14.1343	140.00	49.6169	.00	-.03	.00
	14.0333	145.00	49.6169	.00	.00	.00
	13.9346	150.00	49.6169	.02	.05	.02

MIXTURE NO 30 (Hiza, et al. [14])

0.58692 C₃H₈ + 0.41308 nC₄H₁₀

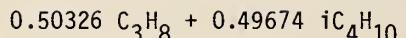
	14.6839	110.00	49.8915	.01	.04	.01
	14.1748	135.00	49.8915	-.03	-.06	-.03
	14.0786	140.00	49.8915	.00	-.02	.00

MIXTURE NO 31 (Hiza, et al., [14])

0.49030 C₃H₈ + 0.50970 iC₄H₁₀

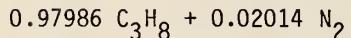
	14.3080	105.00	51.2468	-.01	-.04	-.01
	14.2136	110.00	51.2468	-.01	-.05	.00
	14.1219	115.00	51.2468	.02	-.03	.02
	14.0257	120.00	51.2468	.01	-.04	.01
	13.9300	125.00	51.2468	.00	-.04	.00
	13.8342	130.00	51.2468	-.01	-.04	-.01

MIXTURE NO 32 (Hiza, et al. [14])



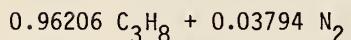
P	D	T	MW	RKM	HS	CELL	CS
	13.9718	125.00	51.0650		.01	-.03	.01
	13.8737	130.00	51.0650		-.01	-.04	-.01

MIXTURE NO 33 (Hiza, et al. [14])



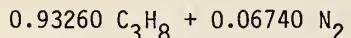
3.567	16.2131	110.00	43.7733		-.01	-.04	-.01
4.712	16.0931	115.00	43.7733		-.02	-.05	-.02

MIXTURE NO 34 (Hiza, et al. [14])



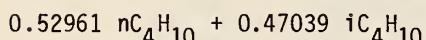
4.955	16.4638	105.00	43.4870		.02	-.03	.01
6.708	16.3410	110.00	43.4870		.00	-.04	.00

MIXTURE NO 35 (Hiza, et al. [14])



8.795	16.7084	105.00	43.0132		.20	.11	.18
6.313	16.8055	100.00	43.0132		.04	-.07	.01

MIXTURE NO 36 (Hiza, et al. [14])



12.6943	125.00	58.1243		-.01	-.03	-.01
12.6133	130.00	58.1243		.01	-.01	.01
12.5271	135.00	58.1243		-.01	-.04	-.01
12.4447	140.00	58.1243		.00	-.02	.00

MIXTURE NO 37 (Hiza and Haynes [15])



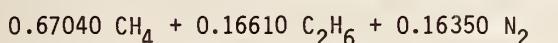
.476	24.9975	105.00	20.1852	.02	.04	.06	.02
.747	24.6696	110.00	20.1852	-.03	.00	.00	-.02
1.119	24.3515	115.00	20.1852	.01	.02	.02	.01
1.616	24.0335	120.00	20.1852	.07	.07	.07	.06

MIXTURE NO 38 (Hiza and Haynes [15])



P	D	T	MW	RKM	HS	CELL	CS
.427	20.5342	110.00	30.0903	- .03	.02	.00	-.03
.636	20.3497	115.00	30.0903	- .01	.03	-.02	-.02
.638	20.3567	115.00	30.0903	.03	.06	.01	.01
.914	20.1786	120.00	30.0903	.12	.10	.04	.06
.914	20.1803	120.00	30.0903	.12	.11	.05	.07
1.280	19.9858	125.00	30.0903	.10	.08	.01	.04

MIXTURE NO 39 (Hiza and Haynes [15])



3.688	25.8587	105.00	20.3301	.57	.01	.28	.00
4.813	25.4598	110.00	20.3301	.64	-.03	.27	-.01
6.181	25.0466	115.00	20.3301	.79	.02	.25	-.05
7.731	24.6359	120.00	20.3301	.70	.42	.31	-.04

MIXTURE NO 40 (Hiza and Haynes [15])



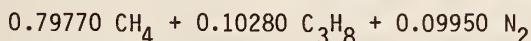
7.190	24.9084	105.00	24.5861	2.06	.01	.61	.07
9.494	24.5350	110.00	24.5861	2.55	-.01	.64	.04
12.372	24.1433	115.00	24.5861	3.72	.24	.65	-.06

MIXTURE NO 41 (Miller and Hiza [25])



.466	24.9239	100.00	20.4561	-.17	.04	.11	-.03
.769	24.4200	108.00	20.4561	-.17	.01	.05	-.04
1.327	23.9676	115.00	20.4561	-.18	-.01	.02	-.05

MIXTURE NO 42 (Hiza and Haynes [15])



2.817	25.3950	105.00	20.1181	.17	.08	.24	-.01
3.620	25.0153	110.00	20.1181	.10	.02	.23	-.05
4.549	24.6465	115.00	20.1181	.08	.08	.33	.01
5.610	24.2667	120.00	20.1181	.02	*	.44	.04

* The hard sphere solution for the density of N₂ failed.

MIXTURE NO 43 (Haynes [9])

0.89071 CH ₄ + 0.04998 nC ₄ H ₁₀ + 0.05931 N ₂							
P	D	T	MW	RKM	HS	CELL	CS
2.400	25.3450	110.00	18.8562	- .23	.00	.01	.03
3.145	24.9440	115.00	18.8562	- .25	- .04	-.02	.01
4.082	24.5383	120.00	18.8562	- .38	*	-.02	.00
5.196	24.1141	125.00	18.8562	- .32	*	-.04	-.04

MIXTURE NO 44 (Miller and Hiza [25])

0.85317 CH ₄ + 0.05077 C ₂ H ₆ + 0.04855 C ₃ H ₈ + 0.04751 nC ₄ H ₁₀							
P	D	T	MW	RKM	HS	CELL	CS
1.038	24.6056	110.04	20.1165	.02	.15	.10	.06
1.464	24.2712	115.00	20.1165	.00	.11	.04	.02

MIXTURE NO 45 (Haynes [9])

0.85133 CH ₄ + 0.05759 C ₂ H ₆ + 0.04808 C ₃ H ₈ + 0.04300 nC ₄ H ₁₀							
P	D	T	MW	RKM	HS	CELL	CS
1.180	24.3243	115.00	20.0092	- .04	.07	.00	-.01
1.700	23.9965	120.00	20.0092	.00	.10	.02	.03
2.374	23.6586	125.00	20.0092	.05	.11	.04	.05
3.232	23.3108	130.00	20.0092	.04	.11	.05	.05
4.301	22.9634	135.00	20.0092	.08	.13	.11	.09

MIXTURE NO 46 (Haynes [9])

0.84566 CH ₄ + 0.07924 C ₂ H ₆ + 0.05060 C ₃ H ₈ + 0.02450 nC ₄ H ₁₀							
P	D	T	MW	RKM	HS	CELL	CS
1.167	24.5569	115.00	19.6051	.00	.04	.00	-.01
1.683	24.2126	120.00	19.6051	.02	.04	-.01	-.01
2.350	23.8698	125.00	19.6051	.08	.07	.03	.03
3.201	23.5204	130.00	19.6051	.12	.11	.09	.07

MIXTURE NO 47 (Haynes [9])

0.86040 CH ₄ + 0.04600 C ₂ H ₆ + 0.04790 C ₃ H ₈ + 0.04570 iC ₄ H ₁₀							
P	D	T	MW	RKM	HS	CELL	CS
1.186	24.2654	115.00	19.9552	- .14	.01	.04	-.03
1.710	23.9371	120.00	19.9552	- .10	.04	.07	.01
2.387	23.5860	125.00	19.9552	- .10	.01	.05	-.01
3.248	23.2331	130.00	19.9552	- .13	-.01	.06	-.02
4.320	22.8637	135.00	19.9552	- .18	-.07	.05	-.07

* The hard sphere solution for the density of N₂ failed.

MIXTURE NO 48 (Haynes [9])

0.85378 CH ₄ + 0.05178 C ₂ H ₆ + 0.04703 C ₃ H ₈ + 0.04741 iC ₄ H ₁₀							
P	D	T	MW	RKM	HS	CELL	CS
1.190	24.2100	115.00	20.0838	- .14	.01	.04	-.03
1.706	23.8779	120.00	20.0838	- .14	.01	.04	-.02
2.379	23.5324	125.00	20.0838	- .13	-.02	.02	-.04
3.238	23.1834	130.00	20.0838	- .15	-.04	.03	-.05

MIXTURE NO 49 (Miller and Hiza [25])

0.85378 CH ₄ + 0.05178 C ₂ H ₆ + 0.04703 C ₃ H ₈ + 0.04741 iC ₄ H ₁₀							
.972	24.5434	110.02	20.0838	- .14	.03	.06	-.01
1.429	24.2154	115.01	20.0838	- .12	.02	.05	-.01

MIXTURE NO 50 (Hiza and Haynes [15])

0.85260 CH ₄ + 0.04830 C ₂ H ₆ + 0.05070 C ₃ H ₈ + 0.04840 N ₂							
1.581	25.8910	105.00	18.7223	.01	-.04	.03	-.07
2.138	25.5081	110.00	18.7223	-.09	-.07	.03	-.07
2.827	25.1224	115.00	18.7223	-.05	-.06	.05	-.06
3.650	24.7283	120.00	18.7223	-.08	*	.08	-.06

MIXTURE NO 51 (Haynes [9])

0.85892 CH ₄ + 0.11532 C ₂ H ₆ + 0.01341 C ₃ H ₈ + 0.00705 nC ₄ H ₁₀ + 0.00530 iC ₄ H ₁₀							
1.185	25.0957	115.00	18.5565	.05	-.02	-.01	-.03
1.706	24.7131	120.00	18.5565	.03	-.06	-.05	-.07
2.372	24.3294	125.00	18.5565	.02	-.07	-.05	-.09
3.225	23.9490	130.00	18.5565	.05	-.02	.01	-.06

MIXTURE NO 52 (Hiza and Haynes [15])

0.85442 CH ₄ + 0.05042 C ₂ H ₆ + 0.04038 C ₃ H ₈ + 0.02901 nC ₄ H ₁₀ + 0.02577 iC ₄ H ₁₀							
.515	24.8775	105.00	20.1885	.09	.25	.25	.16
.818	24.5382	110.00	20.1885	.00	.17	.15	.09
1.190	24.2083	115.00	20.1885	.00	.15	.12	.08
1.695	23.8859	120.00	20.1885	.05	.19	.15	.13

MIXTURE NO 53 (Haynes [9])

0.84558 CH ₄ + 0.08153 C ₂ H ₆ + 0.04778 C ₃ H ₈ + 0.01252 nC ₄ H ₁₀ + 0.01259 iC ₄ H ₁₀							
1.166	24.5586	115.00	19.5838	.01	.05	.04	.01
1.680	24.2180	120.00	19.5838	.04	.07	.05	.03
2.348	23.8688	125.00	19.5838	.08	.08	.08	.04
3.188	23.5154	130.00	19.5838	.10	.10	.12	.07

* The hard sphere solution for the density of N₂ failed.

MIXTURE NO 54 (Haynes [9])

0.81249 CH ₄ + 0.08484 C ₂ H ₆ + 0.04931 C ₃ H ₈ + 0.02708 nC ₄ H ₁₀ + 0.02628 N ₂		P	D	T	MW	RKM	HS	CELL	CS
2.214	24.4562			115.00	20.0706	.01	.01	.03	-.02
2.874	24.1119			120.00	20.0706	.00	*	.04	-.01
3.768	23.7507			125.00	20.0706	.02	*	.02	-.05
4.793	23.3954			130.00	20.0706	-.02	-3.67	.07	-.03

MIXTURE NO 55 (Hiza and Haynes [15])

0.79090 CH ₄ + 0.05600 C ₂ H ₆ + 0.05000 C ₃ H ₈ + 0.04770 nC ₄ H ₁₀ + 0.05540 N ₂		P	D	T	MW	RKM	HS	CELL	CS
1.933	24.8080			105.00	20.9017	.04	.10	.12	.05
2.530	24.4664			110.00	20.9017	.02	.02	.03	-.02

MIXTURE NO 56 (Miller and Hiza [25])

0.79054 CH ₄ + 0.05597 C ₂ H ₆ + 0.04996 C ₃ H ₈ + 0.04762 nC ₄ H ₁₀ + 0.05591 N ₂		P	D	T	MW	RKM	HS	CELL	CS
2.158	24.8354			105.03	20.9029	.15	.21	.23	.15

MIXTURE NO 57 (Haynes [9])

0.80940 CH ₄ + 0.04542 C ₂ H ₆ + 0.05050 C ₃ H ₈ + 0.04667 iC ₄ H ₁₀ + 0.04801 N ₂		P	D	T	MW	RKM	HS	CELL	CS
3.005	24.1487			115.00	20.6355	.02	-.02	-.03	-.04
3.863	23.8075			120.00	20.6355	-.14	*	.00	-.03
4.874	23.4518			125.00	20.6355	-.12	*	.01	-.05
6.125	23.0893			130.00	20.6355	-.18	-5.06	.05	-.08

MIXTURE NO 58 (Hiza and Haynes [15])

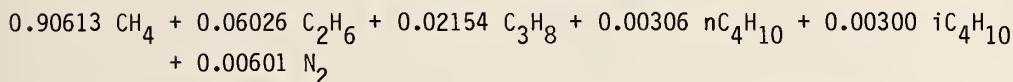
0.80600 CH ₄ + 0.04680 C ₂ H ₆ + 0.04820 C ₃ H ₈ + 0.05000 iC ₄ H ₁₀ + 0.04900 N ₂		P	D	T	MW	RKM	HS	CELL	CS
2.039	24.7803			105.00	20.7423	-.03	.08	.03	.01
2.550	24.4327			110.00	20.7423	-.09	-.02	-.06	-.06
3.135	24.1039			115.00	20.7423	.06	.00	-.02	-.02
3.790	23.7707			120.00	20.7423	-.10	*	-.01	-.02

MIXTURE NO 59 (Miller and Hiza [25])

0.80545 CH ₄ + 0.04671 C ₂ H ₆ + 0.04817 C ₃ H ₈ + 0.04998 iC ₄ H ₁₀ + 0.04969 N ₂		P	D	T	MW	RKM	HS	CELL	CS
2.273	24.7831			105.06	20.7476	.00	.10	.05	.03

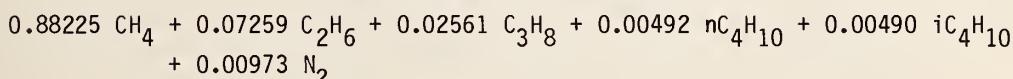
* The hard sphere solution for the density of N₂ failed.

MIXTURE NO 60 (Haynes [9])



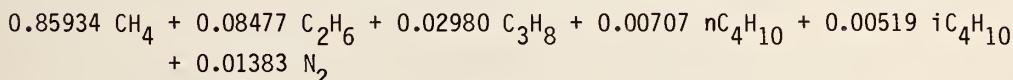
P	D	T	MW	RKM	HS	CELL	CS
1.478	25.3834	115.00	17.8195	.03	-.01	.01	-.01
2.043	24.9894	120.00	17.8195	.08	.00	.05	.02
2.785	24.5702	125.00	17.8195	.04	.00	.03	-.02
3.722	24.1578	130.00	17.8195	.06	-.108	.11	.02

MIXTURE NO 61 (Haynes [9])



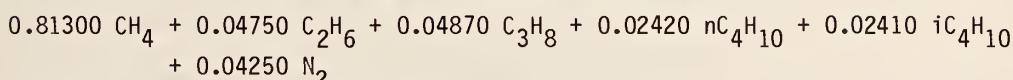
1.639	25.2023	115.00	18.3094	.09	.04	.07	.04
2.247	24.8047	120.00	18.3094	.06	.00	.04	.00
3.022	24.4022	125.00	18.3094	.04	.00	.03	-.03

MIXTURE NO 62 (Haynes [9])



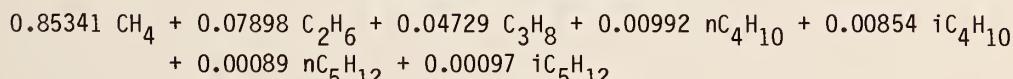
1.812	25.0384	115.00	18.7496	.12	.06	.09	.06
2.441	24.6661	120.00	18.7496	.12	*	.11	.07
3.223	24.2880	125.00	18.7496	.19	.00	.16	.09
4.222	23.8981	130.00	18.7496	.17	-.2.12	.20	.10

MIXTURE NO 63 (Hiza and Haynes [15])



1.834	24.8496	105.00	20.6168	.03	.12	.10	.05
2.807	24.5159	110.00	20.6168	.02	.08	.06	.03
3.384	24.1783	115.00	20.6168	.09	.06	.05	.03
6.039	23.8577	120.00	20.6168	.04	*	.11	.08

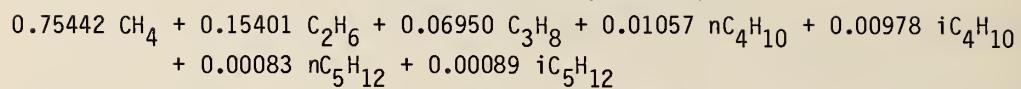
MIXTURE NO 64 (Haynes [9])



.787	25.0063	110.00	19.3588	.01	.02	.02	-.01
1.172	24.6566	115.00	19.3588	.00	-.01	-.01	-.03
1.686	24.3079	120.00	19.3588	.02	.00	.00	-.02
2.351	23.9525	125.00	19.3588	.05	.01	.03	-.01
3.210	23.5883	130.00	19.3588	.06	.02	.06	.00

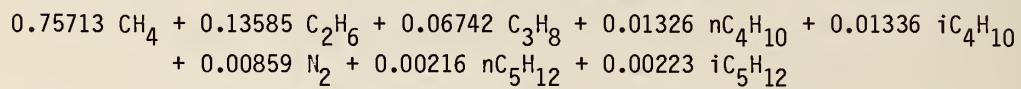
* The hard sphere solution for the density of N₂ failed.

MIXTURE NO 65 (Haynes [9])



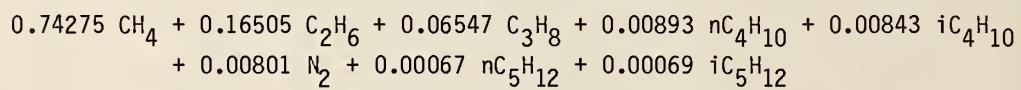
P	D	T	MW	RKM	HS	CELL	CS
.723	24.2529	110.00	21.1060	-.02	.00	-.01	-.04
1.081	23.9619	115.00	21.1060	.04	.04	.03	.01
1.549	23.6535	120.00	21.1060	.07	.04	.02	.01
2.153	23.3351	125.00	21.1060	.04	.01	.00	-.01

MIXTURE NO 66 (Haynes [9])



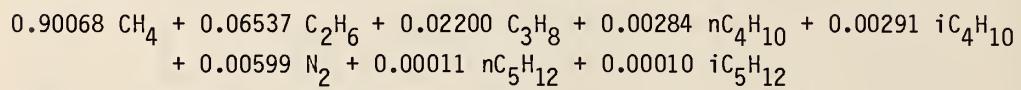
1.155	24.1809	110.00	21.3094	.09	.07	.08	.04
1.595	23.8731	115.00	21.3094	.12	.05	.05	.02
2.155	23.5709	120.00	21.3094	.12	.00	.08	.05
2.873	23.2644	125.00	21.3094	.15	.00	.12	.08
3.744	22.9514	130.00	21.3094	.22	-1.42	.16	.11

MIXTURE NO 67 (Haynes [9])



1.158	24.3141	110.00	21.0976	.02	-.02	.00	-.05
1.584	24.0160	115.00	21.0976	.09	.01	.02	-.01
2.093	23.6937	120.00	21.0976	.03	.00	-.03	-.06
2.853	23.3804	125.00	21.0976	.05	.00	-.01	-.05

MIXTURE NO 68 (Haynes [9])



1.456	25.3600	115.00	17.9026	.05	.01	.03	.01
2.024	24.9656	120.00	17.9026	.08	.00	.06	.03
2.762	24.5450	125.00	17.9026	.03	.00	.03	-.03
3.698	24.1289	130.00	17.9026	.02	-1.13	.07	-.02

Appendix B. Computer Program and Equation Parameters for the Extended Corresponding States Model

The program listings in Appendix F include the extended corresponding states method described in section 2. To use the program in its present form one must make one of the two possible calls to SUBROUTINE PDMIX(P,D,T,X) (lines LNG 1 through LNG 9). The two possible calls are:

```
CALL PDMIX(P,D,T,X)
```

or

```
CALL PMIX(P,D,T,X)
```

When the call to PDMIX(P,D,T,X) is made the input variables are: P (pressure in bars); T (temperature in kelvin); and X which is a matrix of the mole fraction of the components of the mixture in the following order:

- X(1) = mole fraction of methane
- X(2) = mole fraction of ethane
- X(3) = mole fraction of propane
- X(4) = mole fraction of normal butane
- X(5) = mole fraction of isobutane
- X(6) = mole fraction of nitrogen
- X(7) = mole fraction of normal pentane
- X(8) = mole fraction of isopentane

No other components are allowed and if one or more of the above components are absent, a zero should be inserted in the appropriate matrix element. The program then calculates a density and it is returned in the argument list as D (density in moles/liter).

When a call to PMIX(P,D,T,X) is made all of the above are the same except that the roles of P and D are interchanged, i.e., D is an input variable and P is calculated by the program.

The range of the program is 90 to 150 kelvin for the saturated liquid phase of any of the pure components of CH₄, C₂H₆, C₃H₈, nC₄H₁₀, iC₄H₁₀, N₂, nC₄H₁₂ or iC₄H₁₂ or any mixture of those fluids. The program will extrapolate to higher pressures (higher than saturation pressure) but the user is reminded that such a calculation is an extrapolation and should be used with caution.

Other subprograms required:

SUBROUTINE MIX DATA, line LNG 95, Appendix F

SUBROUTINE DATA CH4, line LNG 162, Appendix F

FUNCTION FINDM, line LNG 217, Appendix F

FUNCTION SATL, line LNG 236, Appendix F

SUBROUTINE PROPS, line LNG 248, Appendix F

The equation of state from which Z₀ and G₀ by eqs (1) and (2) may be derived is:

$$\begin{aligned}
P = & \rho RT + \rho^2(N_1 T + N_2 T^{1/2} + N_3 + N_4/T + N_5/T^2) \\
& + \rho^3(N_6 T + N_7 + N_8/T + N_9/T^2) \\
& + \rho^4(N_{10} T + N_{11} + N_{12}/T) + \rho^5(N_{13}) \\
& + \rho^6(N_{14}/T + N_{15}/T^2) + \rho^7(N_{16}/T) \\
& + \rho^8(N_{17}/T + N_{18}/T^2) + \rho^9(N_{19}/T^2) \\
& + \rho^3(N_{20}/T^2 + N_{21}/T^3) \exp(-\gamma\rho^2) \\
& + \rho^5(N_{22}/T^2 + N_{23}/T^4) \exp(-\gamma\rho^2) \\
& + \rho^7(N_{24}/T^2 + N_{25}/T^3) \exp(-\gamma\rho^2) \\
& + \rho^9(N_{26}/T^2 + N_{27}/T^4) \exp(-\gamma\rho^2) \\
& + \rho^{11}(N_{28}/T^2 + N_{29}/T^3) \exp(-\gamma\rho^2) \\
& + \rho^{13}(N_{30}/T^2 + N_{31}/T^3 + N_{32}/T^4) \exp(-\gamma\rho^2)
\end{aligned} \tag{B-1}$$

The computer subroutine PROPS(PP,DD,TT), lines "LNG 248 through LNG 371 in Appendix F" are the FORTRAN statements for eq (B-1) and the derivative $(\partial P/\partial \rho)_T$.

The parameters for eq (B-1) as applied to methane are given in table 2 and in the FORTRAN SUBROUTINE DATA CH4, lines LNG 162 through LNG 216, Appendix F.

The parameters for eqs (5) and (6), (9) and (10) are given in table 3 and 4 and in the FORTRAN SUBROUTINE MIX DATA (IBASE), lines LNG 95 through LNG 161, Appendix F.

Table 2. Methane Coefficients N_i for Eq B-1.

R	=	0.08205616	N_{16}	=	-0.529609525984 $\times 10^{-3}$
γ	=	0.0096	N_{17}	=	0.152264286004 $\times 10^{-4}$
N_1	=	-0.187027997685 $\times 10^{-1}$	N_{18}	=	-0.109952182842 $\times 10^{-1}$
N_2	=	0.103387108009 $\times 10$	N_{19}	=	0.191395549929 $\times 10^{-3}$
N_3	=	-0.155387625619 $\times 10^2$	N_{20}	=	0.386470003746 $\times 10^5$
N_4	=	0.772311478564 $\times 10^3$	N_{21}	=	-0.157930582612 $\times 10^7$
N_5	=	-0.377103300895 $\times 10^5$	N_{22}	=	0.195270144401 $\times 10^3$
N_6	=	0.846818843475 $\times 10^{-3}$	N_{23}	=	0.165996081629 $\times 10^7$
N_7	=	-0.496415884529	N_{24}	=	0.603051146711
N_8	=	0.869909352414 $\times 10^2$	N_{25}	=	0.376485162808 $\times 10^2$
N_9	=	-0.322821592493 $\times 10^5$	N_{26}	=	0.125593680622 $\times 10^{-2}$
N_{10}	=	-0.395843026318 $\times 10^{-4}$	N_{27}	=	-0.343570032513 $\times 10^2$
N_{11}	=	0.266772318035 $\times 10^{-1}$	N_{28}	=	-0.540945094139 $\times 10^{-5}$
N_{12}	=	-0.304010057839 $\times 10$	N_{29}	=	0.185622284663 $\times 10^{-2}$
N_{13}	=	0.191584507536 $\times 10^{-3}$	N_{30}	=	0.770786979245 $\times 10^{-8}$
N_{14}	=	-0.195587933458 $\times 10^{-3}$	N_{31}	=	-0.286868318650 $\times 10^{-5}$
N_{15}	=	0.607479967879 $\times 10$	N_{32}	=	0.372376961647 $\times 10^{-4}$

Table 3. Coefficients for Eqs 5 and 6.

$$\begin{array}{ll}
 n_1 = -0.109495 & n_4 = -4.14192 \\
 n_2 = 0.919454 & n_5 = 0.444850 \\
 n_3 = -4.01525 & \\
 \\
 n_6 = 0.356808 & n_8 = 0.893323 \\
 n_7 = 1.02619 & n_9 = 0.761533
 \end{array}$$

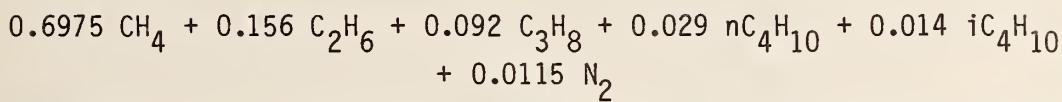
	w	P_c^* (bar)	T_c (K)	V_c (cm^3/mol)	M	Fluid No.
CH_4	0.0109	45.956967	190.555	98.522	16.04303	1
C_2H_6	0.110427	48.60314	305.5	146.2	30.07012	2
C_3H_8	0.154837	42.445123	370.	200.	44.09721	3
$n\text{C}_4\text{H}_{10}$	0.176372	38.295398	425.	251.62	58.1243	4
$i\text{C}_4\text{H}_{10}$	0.150115	36.88998	408.1	263.	58.1243	5
N_2	0.0291791	33.542557	126.2	89.827	28.0134	6
$n\text{C}_5\text{H}_{12}$	0.234320	33.812152	469.6	304.	72.15139	7
$i\text{C}_5\text{H}_{12}$	0.288886	31.988302	460.39	306.	72.15139	8

*Note: The large number of significant figures given for critical pressure is necessary to reproduce the Z_c ($Z_c = P_c V_c / RT_c$) in the least squares fit of the data. P_c 's have been converted to bar from atmospheres.

Table 4. Binary Interaction Coefficients for Eqs 9 and 10.
 Fluid numbers are given in table 3.

i	j	ij	ij	ij	ij	ij	ij
1	1	1.00514	1.01922	1.04243	1.05048	1.01009	1.06
2	1	1.00599	1.01616	1.02369	1.02127	1.02	1.02
3	1	1.00172	1.01140	1.04606	1.01	1.01	1.01
4	1	0.997114	1.20955	1.0	1.0	1.0	1.0
5	1	1	1.13889	1.0	1.0	1.0	1.0
6	1	1	1	1.0	1.0	1.0	1.0
7	1	1	1	1	1	1	1
8							
i	j	ij	ij	ij	ij	ij	ij
1	1	1.01127	0.988608	0.983130	0.986978	0.953430	0.98
2	1	0.999961	0.972223	0.998886	0.939622	0.99	0.99
3	1	0.985547	1.03099	0.912209	0.99	0.99	0.99
4	1	0.976416	0.849200	0.99	0.99	0.99	0.99
5	1	1	0.857310	0.99	0.99	0.99	0.99
6	1	1	1	0.99	0.99	0.99	0.99
7	1	1	1	1	1	1.0	1.0
8							

Table 5. Values for Checking Calculations Using Corresponding States Equations.*



Temperature K	Density in moles/liter	Pressure, bar
95	24.333	1
100	24.067	1.1
105	23.796	1.2

* Included for check purposes only; these values are calculated from the corresponding states model and are not experimental data.

Appendix C. Computer Program and Equation Parameters for the Hard Sphere Model

The program listing that follows is for the hard sphere model described in section 3. To use the program in its present form one must make the following reference to the computer program:

$$\text{DEN} = \text{RODEN}(P, T, X)$$

where DEN is density in moles/liter, P is pressure in bars, T is temperature in kelvin, and X is a matrix of the mole fractions of the components of the mixture in the following order:

- X(1) = mole fraction of methane
- X(2) = mole fraction of ethane
- X(3) = mole fraction of propane
- X(4) = mole fraction of normal butane
- X(5) = mole fraction of isobutane
- X(6) = mole fraction of nitrogen
- X(7) = mole fraction of normal pentane
- X(8) = mole fraction of isopentane

Note: the inclusion of the pentanes is due to Rodosevich and Miller [33] and no optimization of parameters has been included in this work for mixtures with pentane as a component.

The range of the program is 90 to 150 kelvin for the saturated liquid phase of mixtures of CH_4 , C_2H_6 , C_3H_8 , $n\text{C}_4\text{H}_{10}$, $i\text{C}_4\text{H}_{10}$, N_2 , $n\text{C}_5\text{H}_{12}$ and $i\text{C}_5\text{H}_{12}$. The program will calculate densities of any of the pure components but they will be from a different model (i.e., some from an equation of state (CH_4 and N_2) and some from the equations for saturated liquid densities. Therefore in its present form, extrapolation to higher pressures is possible but the reliability of the results is questionable.

Other subprograms required:

SUBROUTINE FM, lines LNG 435 to 529, Appendix F

SUBROUTINE ZERO, lines LNG 617 to 623, Appendix F

FUNCTION FIND V1, lines LNG 530 to 548, Appendix F

FUNCTION FIND G1, lines LNG 549 to 560, Appendix F

FUNCTION EXCESS, lines LNG 561 to 616, Appendix F

FUNCTION FIND M, lines LNG 217 to 234, Appendix F

SUBROUTINE PROPS, lines LNG 248 to 371, Appendix F

SUBROUTINE DATA CH4, lines LNG 162 to 216, Appendix F

SUBROUTINE DATA N2, lines LNG 372 to 415, Appendix F

The parameters for eq (11), section 3 are given in table 6, and in lines LNG 462 through LNG 466 in Appendix F.

The binary interaction parameters j_{ij} and k_{ij} in eqs (15) and (16), section 3 are given in table 7 and in lines LNG 447 through LNG 461 in Appendix F.

Table 6. Coefficients for Eq 11.

Fluid	a_i	s_i	c_i	Fluid No.
CH_4	2.755×10^5	3.676×10^{-8}	1.00	1
C_2H_6	7.773×10^5	4.158×10^{-8}	1.50	2
C_3H_8	14.165×10^5	4.644×10^{-8}	1.67	3
$n\text{C}_4\text{H}_{10}$	22.733×10^5	5.051×10^{-8}	1.83	4
$i\text{C}_4\text{H}_{10}$	21.279×10^5	5.056×10^{-8}	1.79	5
N_2	1.718×10^5	3.546×10^{-8}	1.03	6
$n\text{C}_5\text{H}_{12}$	30.550×10^5	5.389×10^{-8}	1.91	7
$i\text{C}_5\text{H}_{12}$	42.946×10^5	5.706×10^{-8}	2.11	8

$$b_i = (2)(3.14159)(6.025 \times 10^{23})s_i^3/3$$

Table 7. Binary Interaction Parameters for Eqs 15 and 16.

		j _{ij}							k _{ij}								
i	j	1	2	3	4	5	6	7	8	1	2	3	4	5	6	7	8
1	0	-0.388616 × 10 ⁻²		-0.120932 × 10 ⁻¹	-0.231577 × 10 ⁻¹	-0.238349 × 10 ⁻¹	-0.997547 × 10 ⁻²	-0.326 × 10 ⁻¹	0.458 × 10 ⁻¹								
2	0		-0.2162 × 10 ⁻²	-0.400910 × 10 ⁻²	-0.812712 × 10 ⁻²	-0.143976 × 10 ⁻¹	-0.3 × 10 ⁻²	-0.4 × 10 ⁻²									
3	0			0.761571 × 10 ⁻³	-0.383743 × 10 ⁻²	-0.24014 × 10 ⁻¹	0.0	0.0	0.0								
4	0				0.222150 × 10 ⁻²	-0.576043 × 10 ⁻¹	0.0	0.0	0.0								
5	0				0	-0.576043 × 10 ⁻¹	0.0	0.0	0.0								
6						0	0.4 × 10 ⁻¹	0.5 × 10 ⁻¹									
7							0.0	0.0	0.0								
8								0.0	0.0								

Appendix D. Computer Program and Parameters for the Revised
Klosek and McKinley Model

The program listing and tables that follow are for the Revised Klosek and McKinley model described in section 4. The method may be used in two ways. First using the equation:

$$V_{\text{mix}} = \sum x_i V_i - [k_1 + (k_2 - k_1) x_{N_2} / 0.0425] x_{CH_4} \quad (D-1)$$

The V_i , k_1 and k_2 may be obtained from tables 8, 9 and 10 and the volume of the mixture calculated. For example given the mixture of 0.8130 CH_4 + 0.0475 C_2H_6 + 0.0487 C_3H_8 + 0.0242 nC_4H_{10} + 0.0241 iC_4H_{10} + 0.0425 N_2 and a temperature of 105 kelvin.

The $\sum x_i V_i$ and $\sum x_i W_i$ are obtained from table 8.

$$\begin{aligned} \sum x_i V_i &= (.8130)(.037113) + (.0475)(.047267) + (.0487)(.061766) \\ &\quad + (.0242)(.076100) + (.0241)(.077538) + (.0425)(.042565) \\ &= 0.0409453 \end{aligned}$$

$$\begin{aligned} \text{Then } \sum x_i W_i &= (.8130)(16.04303) + (.0475)(30.07012) + (.0487)(44.09721) \\ &\quad + (.0242)(58.1243) + (.0241)(58.1243) + (.0425)(28.0143) \\ &= 20.6168 \text{ the molecular weight of the mixture} \end{aligned}$$

$$\text{from table 9 } k_1 = .697 \times 10^{-3}$$

$$\text{from table 10 } k_2 = .849 \times 10^{-3}$$

plugging all this into eq (D-1) gives

$$V_{\text{mix}} = .040255$$

$$1/V_{\text{mix}} = \rho_{\text{mix}} = 24.842 \text{ moles/liter}$$

This compares to the experimental value of 24.850 (Appendix A, mixture No. 63) to within 0.03%.

The same result may be obtained by using the computer program in the following way:

$$D = \text{FMKM}(T, X)$$

where T is temperature in kelvin and X is a matrix of mole fractions of the components in the following order:

- X(1) = mole fraction of methane
- X(2) = mole fraction of ethane
- X(3) = mole fraction of propane
- X(4) = mole fraction of normal butane
- X(5) = mole fraction of isobutane
- X(6) = mole fraction of nitrogen
- X(7) = mole fraction of normal pentane
- X(8) = mole fraction of isopentane

for the example: T = 105., X(1) = .8130, X(2) = .0475, X(3) = .0487, X(4) = .0242, X(5) = .0241 and X(6) = .0425.

Other subprograms required:

FUNCTION VIDEAL, lines LNG 709 through 732, Appendix F

FUNCTION SAT, lines LNG 733 through 762, Appendix F

Table 8. Volumes of Saturated Liquid of the Pure Components in
Liters/mole.

T, K	CH ₄	C ₂ H ₆	C ₃ H ₈	nC ₄ H ₁₀	iC ₄ H ₁₀	N ₂	nC ₅ H ₁₂	iC ₅ H ₁₂
90.	.035441	.046081	.060461	.074708	.076084	.037543	.089173	.089243
92.	.035649	.046235	.060632	.074891	.076274	.038081	.089379	.089454
94.	.035861	.046390	.060804	.075075	.076466	.038650	.089586	.089666
96.	.036077	.046547	.060977	.075259	.076659	.039254	.089793	.089878
98.	.036298	.046704	.061151	.075445	.076853	.039897	.090000	.090091
100.	.036524	.046863	.061325	.075631	.077047	.040586	.090208	.090304
102.	.036755	.047023	.061501	.075818	.077243	.041327	.090416	.090518
104.	.036992	.047185	.061677	.076006	.077440	.042128	.090624	.090733
106.	.037234	.047348	.061855	.076194	.077637	.043002	.090833	.090948
108.	.037481	.047512	.062033	.076384	.077836	.043963	.091042	.091163
110.	.037735	.047678	.062212	.076574	.078035	.045031	.091252	.091379
112.	.037995	.047845	.062392	.076765	.078236	.046231	.091462	.091596
114.	.038262	.048014	.062574	.076957	.078438	.047602	.091673	.091814
116.	.038536	.048184	.062756	.077150	.078640	.049179	.091884	.092032
118.	.038817	.048356	.062939	.077344	.078844	.050885	.092095	.092251
120.	.039106	.048529	.063124	.077539	.079049	.052714	.092307	.092470
122.	.039404	.048704	.063309	.077734	.079255	.054679	.092520	.092690
124.	.039710	.048881	.063496	.077931	.079462	.056797	.092733	.092911
126.	.040025	.049059	.063684	.078128	.079671	.059085	.092947	.093133
128.	.040350	.049239	.063873	.078327	.079880	.061565	.093161	.093355
130.	.040685	.049421	.064063	.078526	.080091	.064263	.093376	.093578
molecular weight*		16.04303	30.07012	44.09721	58.1243	28.0143	72.15139	72.15139

Table 9. Correction Factor $k_1 \times 10^3$.

T/W	16	17	18	19	20	21	22	23	24	25
90	-.005	.120	.220	.340	.430	.515	.595	.660	.725	.795
95	-.006	.135	.260	.380	.500	.590	.665	.740	.810	.885
100	-.007	.150	.300	.425	.575	.675	.755	.830	.910	.990
105	-.007	.165	.340	.475	.635	.735	.840	.920	1.045	1.120
110	-.008	.180	.375	.535	.725	.835	.950	1.055	1.155	1.245
115	-.009	.220	.440	.610	.810	.945	1.065	1.180	1.280	1.380
120	-.01	.250	.500	.695	.920	1.055	1.205	1.330	1.450	1.550
125	-.013	.295	.590	.795	1.035	1.210	1.385	1.525	1.640	1.750
130	-.015	.345	.700	.920	1.200	1.370	1.555	1.715	1.860	1.990
135	-.017	.400	.825	1.060	1.390	1.590	1.800	1.950	2.105	2.272

Table 10. Correction Factor $k_2 \times 10^3$.

T/W	16	17	18	19	20	21	22	23	24	25
90	-.004	.10	.22	.35	.50	.60	.69	.78	.86	.95
95	-.005	.12	.28	.43	.59	.71	.83	.94	1.05	1.14
100	-.007	.16	.34	.49	.64	.79	.94	1.08	1.17	1.27
105	-.01	.24	.42	.61	.75	.91	1.05	1.19	1.33	1.45
110	-.015	.32	.59	.77	.92	1.07	1.22	1.37	1.52	1.71
115	-.024	.41	.72	.95	1.15	1.22	1.3	1.45	1.65	2.00
120	-.032	.60	.91	1.23	1.43	1.63	1.85	2.08	2.30	2.45
125	-.043	.71	1.13	1.48	1.73	1.98	2.23	2.48	2.75	2.90
130	-.058	.95	1.46	1.92	2.20	2.42	2.68	3.00	3.32	3.52
135	-.075	1.30	2.00	2.40	2.60	3.00	3.40	3.77	3.99	4.23

Appendix E. Computer Program for the Cell Model

The program listings for the cell model start at line LNG 763 and continue on to the end of Appendix F. As is mentioned in section 4, no details of the model are given here only the program listing. To use the program in its present form one must make the following reference to the computer program:

CALL ECKNON(P,D,T,X)

where P is input pressure in bars, D is the output density in moles/liter, T is the input temperature in kelvins and X is a matrix of the mole fractions of the components of the mixture in the following order:

- X(1) = mole fraction of methane
- X(2) = mole fraction of ethane
- X(3) = mole fraction of propane
- X(4) = mole fraction of normal butane
- X(5) = mole fraction of isobutane
- X(6) = mole fraction of nitrogen
- X(7) = mole fraction of normal pentane
- X(8) = mole fraction of isopentane

Appendix F. Computer Programs

Listing of computer programs for all four models. See the sections on the individual models for a list of subprograms needed for each model.

The programs are written in FORTRAN IV and are operational on a CDC 6600 computer.

```

SUBROUTINE PDMIX(P,D,T,X) LNG 1
C FOR A CALL TO PDMIX, P,T AND X ARE INPUT. P IS IN BAR,T IS IN LNG 2
C KELVIN AND D IS OUTPUT IN THE UNITS OF MOLES/LITER LNG 3
C FOR A CALL TO P MIX, D,T AND X ARE INPUT AND P IS OUTPUT, THE LNG 4
C UNITS ARE THE SAME LNG 5
C THE X MATRIX MUST CONTAIN THE MOL FRACTION OF THE ALLOWABLE FLUIDS LNG 6
C IN THE FOLLOWING ORDER,1=C1,2=C2,3=C3,4=NC4,5=IC4,6=N2,7=NC5,8=IC5 LNG 7
C PLACE A ZERO IN THE ELEMENTS OF X WHERE THAT PARTICULAR GAS IS NOT LNG 8
C PRESENT LNG 9
DIMENSION ZATA(10,10),ATA(10,10),TC(10),VC(10),ZC(10),AC(10),W(10) LNG 10
1,PC(10),CF(9) LNG 11
DIMENSION THETA(10,10),TH(10,10),PHI(10,10),PH(10,10),F(10,10),FH LNG 12
1(10,10),H(10,10),HH(10,10),VR(10,10),TR(10,10),X(10) LNG 13
COMMON/DATA M/ZATA,ATA,TC,VC,W,TC0,VCO,AC0,ZC0,RR,R,OMEGO,AC,ZC,N LNG 14
1,PC,CF LNG 15
DATA(IE=0) LNG 16
D=0.0 $ PI=P/1.01325 LNG 17
GO TO 4 LNG 18
ENTRY PMIX LNG 19
P=0.0 LNG 20
IP=1 LNG 21
GO TO 5 LNG 22
4 IP=0 LNG 23
5 CONTINUE LNG 24
IF(IE.GT.0)GO TO 6 LNG 25
IE=1 LNG 26
CALL DATA CH4 LNG 27
IBASE=L=1 LNG 28
CALL MIX DATA(IBASE) LNG 29
6 CONTINUE LNG 30
DO 1 I=1,N LNG 31
F(I,I)=H(I,I)=1. LNG 32
THETA(I,I)=1. LNG 33
1 PHI(I,I)=1. LNG 34
DO 30 J=1,30 LNG 35
HX=FXHX=0.0 LNG 36
DO 10 I=1,N LNG 37
FH(I,I)=F(I,I) LNG 38
HH(I,I)=H(I,I) LNG 39
IF(X(I).LT..0001)GO TO 10 LNG 40
F(I,I)=(TC(I)/TC(L))*THETA(I,I) LNG 41
H(I,I)=(VC(I)/VC(L))*PHI(I,I) LNG 42
10 CONTINUE LNG 43
DO 11 IA=1,N LNG 44
DO 11 IB=1,N LNG 45
IF(X(IA).LT..0001)GO TO 11 LNG 46
IF(X(IB).LT..0001)GO TO 11 LNG 47
FAB=ZATA(IA,IB)*(F(IA,IA)*F(IB,IB))**.5 LNG 48
HAB=ATA(IA,IB)*(5*H(IA,IA)**(1./3.)+5*H(IB,IB)**(1./3.))**3 LNG 49
HX=HX+X(IA)*X(IB)*HAB LNG 50
FXHX=FXHX+X(IA)*X(IB)*HAB*FAB LNG 51
11 CONTINUE LNG 52
FX=FXHX/HX LNG 53
PRO=PI*HX/FX LNG 54
TRO=T/FX LNG 55
DEN=D*HX LNG 56
IF(IP.EQ.1)GO TO 8 LNG 57
DD=SATL(TRO)*1000.+1. LNG 58
9 DEN=FIND M(PRO,TRO,DD) LNG 59
8 CONTINUE LNG 60
IF(DEN.LE.0.0) GO TO 33 LNG 61
VRO=1000./DEN LNG 62
DO 12 I=1,N LNG 63

```

```

IF(X(I).LT..0001)GO TO 12                      LNG  64
VR(I,I)=VRO*PHI(I,I)/VCO                      LNG  65
TR(I,I)=TRO*THETA(I,I)/TCO                      LNG  66
IF(VR(I,I).GT.2.)VR(I,I)=2.                      LNG  67
IF(VR(I,I).LT..5)VR(I,I)=.5                     LNG  68
TH(I,I)=THETA(I,I)                             LNG  69
THETA(I,I)=1.+(AC(I)-OMEGO)*(CF(1)-CF(2)*ALOG(TR(I,I))+(CF(3)-CF(4
1)/TR(I,I))*(VR(I,I)-CF(5)))                  LNG  70
PH(I,I)=PHI(I,I)                               LNG  71
PHI(I,I)=(1.+(AC(I)-OMEGO)*(CF(6)*(VR(I,I)-CF(7))-CF(8)*(VR(I,I)
1-CF(9)*ALOG(TR(I,I)))))*ZCO/ZC(I)          LNG  72
LNG  73
LNG  74
12 CONTINUE                                     LNG  75
DO 13 I=1,N                                    LNG  76
IF(X(I).LT..0001)GO TO 13                      LNG  77
IF(ABS ((FH(I,I)-F(I,I))/F(I,I)).GT..001)GO TO 30    LNG  78
IF(ABS ((HH(I,I)-H(I,I))/H(I,I)).GT..001)GO TO 30    LNG  79
IF(ABS ((TH(I,I)-THETA(I,I))/THETA(I,I)).GT..001)GO TO 30    LNG  80
IF(ABS ((PH(I,I)-PHI(I,I))/PHI(I,I)).GT..001)GO TO 30    LNG  81
LNG  82
13 CONTINUE                                     LNG  83
GO TO 31                                       LNG  84
30 CONTINUE                                     LNG  85
33 PRINT 100,P,DEN,T                          LNG  86
100 FORMAT(* ITTERATION FAILED AT*,3F10.4)
STOP                                         LNG  87
31 D=DEN/HX                                    LNG  88
THC=THETA(6,6)                                LNG  89
IF(IP.EQ.0)GO TO 32                          LNG  90
CALL PRESS(P,DEN,TR0)                         LNG  91
PI=1.01325*P*FX/HX                           LNG  92
32 RETURN                                      LNG  93
END                                           LNG  94
SUBROUTINE MIX DATA(IBASE)                    LNG  95
DIMENSION ZATA(10,10),ATA(10,10),TC(10),VC(10),ZC(10),AC(10),W(10) LNG  96
1,PC(10),CF(9)                                LNG  97
COMMON/DATA M/ZATA,ATA,TC,VC,W,TCO,VCO,ACO,ZCO,RR,R,OMEGO,AC,ZC,N LNG  98
1,PC,CF                                         LNG  99
DATA(PC=45.356,47.96757,41.89008,37.79462,36.40758,33.10393) LNG 100
DATA(TC=190.555,305.5,370.,425.,408.1,126.2)      LNG 101
DATA(W=16.04303,30.07012,44.09721,58.1243,58.1243,28.0134) LNG 102
DATA(VC=98.522,146.2,200.,251.62,263.00,89.827)     LNG 103
DATA(AC=.0109,.110427,.154837,.176372,.150115,.0291791) LNG 104
DATA(R=82.05606),(RR=8.3144),(N=6)                LNG 105
DATA(CF=-.109495,.919454,-4.01525,-4.14192,.444850,.356808, LNG 106
11.02619,.893323,.761533)                      LNG 107
DATA(ATA(1,2)=1.00514),(ATA(1,3)=1.01922),(ATA(1,4)=1.04243), LNG 108
1(ATA(1,5)=1.05048),(ATA(1,6)=1.01009),           LNG 109
A(ATA(2,3)=1.00599),(ATA(2,4)=1.01616),           LNG 110
2(ATA(2,5)=1.02369),(ATA(2,6)=1.02127),(ATA(3,4)=1.00172), LNG 111
3(ATA(3,5)=1.01140),(ATA(3,6)=1.04606),(ATA(4,5)=.997114), LNG 112
4(ATA(4,6)=1.20955),(ATA(5,6)=1.13889)          LNG 113
DATA(ZATA(1,2)=1.01127),(ZATA(1,3)=.988608),(ZATA(1,4)=.983130), LNG 114
1(ZATA(1,5)=.986978),(ZATA(1,6)=.953430),(ZATA(2,3)=.999961), LNG 115
2(ZATA(2,4)=.972223),(ZATA(2,5)=.998886),(ZATA(2,6)=.939622), LNG 116
3(ZATA(3,4)=.985547),(ZATA(3,5)=1.03099),(ZATA(3,6)=.912209), LNG 117
4(ZATA(4,5)=.976416),(ZATA(4,6)= .849200),(ZATA(5,6)=.857310) LNG 118
L=IBASE                                         LNG 119
N=8                                            LNG 120
N1=N-1                                         LNG 121
PC(7)=33.37                                    LNG 122
PC(8)=31.57                                    LNG 123
TC(7)=469.6                                     LNG 124
TC(8)=460.39                                    LNG 125
VC(7)=304.                                         LNG 126

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VC(8)=306.                                              LNG 127
W(7)=72.15139                                         LNG 128
W(8)=72.15139                                         LNG 129
AC(7)=-.234320                                         LNG 130
AC(8)=.288886                                         LNG 131
ATA(1,7)=ATA(1,8)=1.06                                LNG 132
ATA(2,7)=ATA(2,8)=1.02                                LNG 133
ATA(3,7)=ATA(3,8)=1.01                                LNG 134
ATA(4,7)=ATA(4,8)=1.                                   LNG 135
ATA(5,7)=ATA(5,8)=1.                                   LNG 136
ATA(6,7)=ATA(6,8)=1.                                   LNG 137
ATA(7,8)=1.                                             LNG 138
ZATA(1,7)=ZATA(1,8)=.98                               LNG 139
ZATA(2,7)=ZATA(2,8)=.99                               LNG 140
ZATA(3,7)=ZATA(3,8)=.99                               LNG 141
ZATA(4,7)=ZATA(4,8)=.99                               LNG 142
ZATA(5,7)=ZATA(5,8)=.99                               LNG 143
ZATA(6,7)=ZATA(6,8)=.99                               LNG 144
ZATA(7,8)=1.                                           LNG 145
DO 3 J=1,N1                                           LNG 146
J1=J+1                                               LNG 147
DO 3 K=J1,N                                         LNG 148
ZATA(K,J)=ZATA(J,K)                                 LNG 149
3 ATA(K,J)=ATA(J,K)                                 LNG 150
DO 4 I=1,N                                         LNG 151
4 ATA(I,I)=ZATA(I,I)=1.                            LNG 152
PCO=PC(L)                                           LNG 153
VCO=VC(L)                                           LNG 154
TCO=TC(L)                                           LNG 155
OMEGO=AC(L)                                         LNG 156
ZCO=PCO*VCO/TCO/R                                LNG 157
DO 5 I=1,N                                         LNG 158
5 ZC(I)=PC(I)*VC(I)/TC(I)/R                         LNG 159
RETURN                                              LNG 160
END                                                 LNG 161
SUBROUTINE DATA CH4                                  LNG 162
C INITIALIZES THE EQUATION OF STATE CONSTANTS TO METHANE LNG 163
DIMENSION G(32),VP(9),GI(11)                         LNG 164
COMMON/DATA/G,R,GAMMA,VP,DTP                         LNG 165
DIMENSION A(10)                                         LNG 166
COMMON/SATC/A                                         LNG 167
R=.08205616                                         LNG 168
GAMMA=-.0096                                         LNG 169
A(1)=190.555                                         LNG 170
A(2)=10.23                                           LNG 171
A(3)=18.404156472                                    LNG 172
A(4)=7.3498921512                                    LNG 173
A(5)=-1.4313160833                                 LNG 174
A(6)=A(7)=0.0                                         LNG 175
G( 1)=-.187027997685E-01                           LNG 176
G( 2)= .103387108009E+01                           LNG 177
G( 3)=-.155387625619E+02                           LNG 178
G( 4)= .772311478564E+03                           LNG 179
G( 5)=-.377103300895E+05                           LNG 180
G( 6)= .846818843475E-03                           LNG 181
G( 7)=-.496415884529E+00                           LNG 182
G( 8)= .869909352414E+02                           LNG 183
G( 9)=-.322821592493E+05                           LNG 184
G(10)=-.395843026318E-04                           LNG 185
G(11)= .266772318035E-01                           LNG 186
G(12)=-.304010057839E+01                           LNG 187
G(13)= .191584507536E-03                           LNG 188
G(14)=-.195587933458E-03                           LNG 189

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G(15)= .607479967879E+01 LNG 190
G(16)=-.529609525984E-03 LNG 191
G(17)= .152264286004E-04 LNG 192
G(18)=-.109952182842E-01 LNG 193
G(19)= .191395549929E-03 LNG 194
G(20)= .386470003746E+05 LNG 195
G(21)=-.157930582612E+07 LNG 196
G(22)= .195270144401E+03 LNG 197
G(23)= .165996081629E+07 LNG 198
G(24)= .603051146711E+00 LNG 199
G(25)= .376485162808E+02 LNG 200
G(26)= .125593680622E-02 LNG 201
G(27)=-.343570032513E+02 LNG 202
G(28)=-.540945094139E-05 LNG 203
G(29)= .185622284663E-02 LNG 204
G(30)= .770786979245E-08 LNG 205
G(31)=-.286868318650E-05 LNG 206
G(32)= .372376961647E-04 LNG 207
VP(1)=4.77748580 LNG 208
VP(2)=1.76065363 LNG 209
VP(3)=-.56788894 LNG 210
VP(4)=1.32786231 LNG 211
VP(5)=1.5 LNG 212
VP(6)=.1159 LNG 213
VP(7)=90.68 LNG 214
VP(8)=190.555 LNG 215
END LNG 216
FUNCTION FIND M(P,T,D) LNG 217
C SOLVES THE EQUATION OF STATE OF METHANE FOR DENSITY GIVEN P AND T LNG 218
DD=D LNG 219
TT=T LNG 220
DO 10 I=1,50 LNG 221
CALL PRESS(PP,DD,TT) LNG 222
P2=PP LNG 223
IF(ABS (P-P2)-1.E-7*P)20,20,1 LNG 224
1 CALL DPDD(PP,DD,TT) LNG 225
DP=PP LNG 226
CORR=(P2-P)/DP LNG 227
IF(ABS (CORR)-1.E-7*DD)20,20,10 LNG 228
10 DD=DD-CORR LNG 229
FIND M=0 LNG 230
RETURN LNG 231
20 FIND M=DD LNG 232
RETURN LNG 233
END LNG 234
FUNCTION SATL(T) LNG 235
C CALCULATES THE SATURATED LIQUID DENSITY OF METHANE LNG 236
DIMENSION A(10) LNG 237
COMMON/SATC/A LNG 238
IF(T.GT.A(1))GO TO 1 LNG 239
X=(1.-T/A(1)) LNG 240
SATL=A(2)+A(3)*X**(.35)+A(4)*X+A(5)*X**(.3.)+A(6)*X**(.3./.3.) LNG 241
1)+A(7)*X**2 LNG 242
SATL=SATL/1000. LNG 243
RETURN LNG 244
1 SATL=1.E20 LNG 245
RETURN LNG 246
END LNG 247
SUBROUTINE PROPS(PP,DD,TT) LNG 248
C EQUATION OF STATE FOR METHANE AND NITROGEN LNG 249
DIMENSION X(33) LNG 250
DIMENSION B(33),G(32) LNG 251
EQUIVALENCE (B,X) LNG 252

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COMMON/DATA/G,R,GAMMA	LNG 253
DATA(ID=1)	LNG 254
DATA(IZ=1)	LNG 255
1 CONTINUE	LNG 256
D=DD	LNG 257
P=PP	LNG 258
T=TT	LNG 259
GM=GAMMA	LNG 260
D2=D*D	LNG 261
D3=D2*D	LNG 262
D4=D3*D	LNG 263
D5=D4*D	LNG 264
D6=D5*D	LNG 265
D7=D6*D	LNG 266
D8=D7*D	LNG 267
D9=D8*D	LNG 268
D10=D9*D	LNG 269
D11=D10*D	LNG 270
D12=D11*D	LNG 271
D13=D12*D	LNG 272
TS=SQRT (T)	LNG 273
T2=T*T	LNG 274
T3=T2*T	LNG 275
T4=T3*T	LNG 276
T5=T4*T	LNG 277
F=EXP (GM*D2)	LNG 278
GO TO (100,200),K	LNG 279
ENTRY PRESS	LNG 280
K=1	LNG 281
GO TO 1	LNG 282
100 CONTINUE	LNG 283
B(1)=D2*T	LNG 284
B(2)=D2*TS	LNG 285
B(3)=D2	LNG 286
B(4)=D2/T	LNG 287
B(5)=D2/T2	LNG 288
B(6)=D3*T	LNG 289
B(7)=D3	LNG 290
B(8)=D3/T	LNG 291
B(9)=D3/T2	LNG 292
B(10)=D4*T	LNG 293
B(11)=D4	LNG 294
B(12)=D4/T	LNG 295
B(13)=D5	LNG 296
B(14)=D6/T	LNG 297
B(15)=D6/T2	LNG 298
B(16)=D7/T	LNG 299
B(17)=D8/T	LNG 300
B(18)=D8/T2	LNG 301
B(19)=D9/T2	LNG 302
B(20)=D3*F/T2	LNG 303
B(21)=D3*F/T3	LNG 304
B(22)=D5*F/T2	LNG 305
B(23)=D5*F/T4	LNG 306
B(24)=D7*F/T2	LNG 307
B(25)=D7*F/T3	LNG 308
B(26)=D9*F/T2	LNG 309
B(27)=D9*F/T4	LNG 310
B(28)=D11*F/T2	LNG 311
B(29)=D11*F/T3	LNG 312
B(30)=D13*F/T2	LNG 313
B(31)=D13*F/T3	LNG 314
B(32)=D13*F/T4	LNG 315

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102 P=0 LNG 316
    DO 101 I=1,32 LNG 317
101  P=P+B(I)*G(I) LNG 318
    P=P+R*D*T LNG 319
    PP=P LNG 320
    RETURN LNG 321
    ENTRY DPDD LNG 322
    K=2 LNG 323
    GO TO 1 LNG 324
200 CONTINUE LNG 325
    F1=2.00*F*GM*D LNG 326
    F21=3.000*F*D2 +F1*D3 LNG 327
    F22=5.000*F*D4 +F1*D5 LNG 328
    F23=7.000*F*D6 +F1*D7 LNG 329
    F24=9.000*F*D8 +F1*D9 LNG 330
    F25=11.00*F*D10+F1*D11 LNG 331
    F26=13.00*F*D12+F1*D13 LNG 332
    B( 1)=2.00*D*T LNG 333
    B( 2)=2.00*D*TS LNG 334
    B( 3)=2.00*D LNG 335
    B( 4)=2.00*D/T LNG 336
    B( 5)=2.00*D/T2 LNG 337
    B( 6)=3.00*D2*T LNG 338
    B( 7)=3.00*D2 LNG 339
    B( 8)=3.00*D2/T LNG 340
    B( 9)=3.00*D2/T2 LNG 341
    B(10)=4.00*D3*T LNG 342
    B(11)=4.00*D3 LNG 343
    B(12)=4.00*D3/T LNG 344
    B(13)=5.00*D4 LNG 345
    B(14)=6.00*D5/T LNG 346
    B(15)=6.00*D5/T2 LNG 347
    B(16)=7.00*D6/T LNG 348
    B(17)=8.00*D7/T LNG 349
    B(18)=8.00*D7/T2 LNG 350
    B(19)=9.00*D8/T2 LNG 351
    B(20)=F21/T2 LNG 352
    B(21)=F21/T3 LNG 353
    B(22)=F22/T2 LNG 354
    B(23)=F22/T4 LNG 355
    B(24)=F23/T2 LNG 356
    B(25)=F23/T3 LNG 357
    B(26)=F24/T2 LNG 358
    B(27)=F24/T4 LNG 359
    B(28)=F25/T2 LNG 360
    B(29)=F25/T3 LNG 361
    B(30)=F26/T2 LNG 362
    B(31)=F26/T3 LNG 363
    B(32)=F26/T4 LNG 364
202 P=0 LNG 365
    DO 201 I=1,32 LNG 366
201  P=P+B(I)*G(I) LNG 367
    P=P+R*T LNG 368
    PP=P LNG 369
    RETURN LNG 370
    END LNG 371
    SUBROUTINE DATA N2 LNG 372
    INITIALIZES THE EQUATION OF STATE CONSTANTS TO NITROGEN LNG 373
    DIMENSION G(32),VP(9),GI(11) LNG 374
    COMMON/DATA/G,R,GAMMA,VP,DTP LNG 375
    R=8.20539E-2 LNG 376
    GAMMA=-.0056 LNG 377
    G( 1)= 0.136224769272827E-02 LNG 378

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G( 2)= 0.107032469908591E 00 LNG 379
G( 3)= -0.243900721871413E 01 LNG 380
G( 4)= 0.341007449376470E 02 LNG 381
G( 5)= -0.422374309466167E 04 LNG 382
G( 6)= 0.105098600246494E-03 LNG 383
G( 7)= -0.112594826522081E-01 LNG 384
G( 8)= 0.142600789270907E-03 LNG 385
G( 9)= 0.184698501609007E 05 LNG 386
G(10)= 0.811140082588776E-07 LNG 387
G(11)= 0.233011645038006E-02 LNG 388
G(12)= -0.507752586350986E 00 LNG 389
G(13)= 0.485027881931214E-04 LNG 390
G(14)= -0.113656764115364E-02 LNG 391
G(15)= -0.707430273540575E 00 LNG 392
G(16)= 0.751706648852680E-04 LNG 393
G(17)= -0.111614119537424E-05 LNG 394
G(18)= 0.368796562233495E-03 LNG 395
G(19)= -0.201317691347729E-05 LNG 396
G(20)= -0.169717444755949E 05 LNG 397
G(21)= -0.119719240044192E 06 LNG 398
G(22)= -0.975218272038281E 02 LNG 399
G(23)= 0.554639713151823E 05 LNG 400
G(24)= -0.179920450443470E 00 LNG 401
G(25)= -0.256582926077184E 01 LNG 402
G(26)= -0.413707715090789E-03 LNG 403
G(27)= -0.256245415300293E 00 LNG 404
G(28)= -0.124222373740063E-06 LNG 405
G(29)= 0.103556535840165E-04 LNG 406
G(30)= -0.538699166558303E-09 LNG 407
G(31)= -0.757415412839596E-08 LNG 408
G(32)= 0.585367172069521E-07 LNG 409
VP(1)=5.1113192094 $ VP(2)=6.482667539E-1 LNG 410
VP(3)=-1.5108730916E-1 $ VP(4)=7.4028493342E-1 LNG 411
VP(5)=1.5 $ VP(6)=.123 $ VP(7)=63.15 $ VP(8)=126.26 LNG 412
VP(9)=0.0 LNG 413
DTP=31.0 LNG 414
RETURN $ END LNG 415
FUNCTION VPN(TT) LNG 416
C CALCULATES THE VAPOR PRESSURE OF BOTH METHANE AND NITROGEN LNG 417
DIMENSION G(32),VP(9) LNG 418
COMMON/DATA/G,R,GAMMA,VP,DTP LNG 419
T=TT LNG 420
X=(1.-VP(7)/T)/(1.-VP(7)/VP(8)) LNG 421
VPN=VP(6)*EXP(VP(1)*X+VP(2)*X*X+VP(3)*X**3+VP(9)*X**4+VP(4)*X*  
1(1.-X)**VP(5)) LNG 422
RETURN LNG 423
END LNG 424
FUNCTION RODEN(P,T,X) LNG 425
C THE HARD SPHERE MODEL, SEE SUBROUTINE FM FOR THE ARGUMENT LIST LNG 427
DIMENSION X(10) LNG 428
CALL FM(P,T,X,V,G) LNG 429
D=28. LNG 430
V=V+V IDEL(P,D,T,X) LNG 431
RODEN=1000./V LNG 432
RETURN LNG 433
END LNG 434
SUBROUTINE FM(Q,T,X,V9,G9) LNG 435
C PREDICTION OF EXCESS PROPERTIES WITH LHW POTENTIAL, COMP 1-METH LNG 436
C ANE, COMP 2-ETHANE, COMP 3-PROPANE, COMP 4 N-BUTANE, COMP 5-I-BUT LNG 437
C ANE, COMP 6-NITROGEN, COMP 7-NORMAL PENTANE, COMP 8-ISOPENTANE LNG 438
C ADV CRYO ENGR. VOL. 19 (1973)-REPROGRAMED BY R. MCCARTY, 2/22/74 LNG 439
C ARGUMENTS ARE X-MOLE FRACTIONS, T-TEMPERATURE, Q-PRESSURE, V9-EXCESS LNG 440
C VOLUME, G9-EXCESS GIBBS ENERGY, FIRST THREE ARE INPUT, LAST TWO ARE LNG 441

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C OUTPUT, INPUT IS IN KELVINS AND BAR, LNG 442
    DIMENSION A(8),B(8) ,S(8),V(8),G(8),C(8,8),D(8,8),K(8,8), LNG 443
    1J(8,8),E(8),Y(8),O(8),X(10) LNG 444
    TYPE REAL J,K,N1 LNG 445
    DATA(KEY=1) LNG 446
    DATA(J(1,2)=-.00388616),(J(1,3)=-.0120932),(J(1,4)=-.0231577), LNG 447
    A(J(1,5)=-.0238349),(J(1,6)=-.00997547),(J(1,7)=-.0326),(J(1,8)= LNG 448
    B-.0458),(J(2,4)=-.00400910),(J(2,5)=-.00812712),(J(2,6)=-.0143976) LNG 449
    C,(J(2,7)=-.003),(J(2,8)=-.004),(J(3,4)=+.0007615710),(J(3,5)=-.003 LNG 450
    D83743),(J(3,6)=-.024014),(J(3,7)=0.0),(J(3,8)=0.0),(J(4,5)=.002221 LNG 451
    E50),(J(4,7)=.0),(J(4,8)=.0),(J(5,6)=-.0576043),(J(4,6)=-.0576043), LNG 452
    F(J(5,7)=.0),(J(5,8)=0.0),(J(6,7)=-.04),(J(6,8)=-.05),(J(7,8)=.0) LNG 453
    G ,(J(2,3)=-.002162) LNG 454
    DATA(K(1,2)=.00298830),(K(1,3)=.0597378),(K(1,4)=.110893),( LNG 455
    AK(1,5)=.100298),(K(1,6)=.0197290),(K(1,7)=.14),(K(1,8)=.1745), LNG 456
    B(K(2,4)=.0677703),(K(2,5)=.0346632),(K(2,6)=.0529034),(K(2,7)= LNG 457
    C.02),(K(2,8)=.03),(K(3,4)=.0249291),(K(3,5)=-.00838212),(K(4,5)=.0 LNG 458
    D199213),(K(4,6)=.154365),(K(4,7)=.0),(K(4,8)=.0),(K(5,6)=.154365) LNG 459
    E,(K(5,7)=.0),(K(5,8)=.0),(K(6,7)=.15),(K(6,8)=.18),(K(7,8)=0.0) LNG 460
    F,(K(2,3)=.014527),(K(3,6)=.14719),(K(3,7)=0.0),(K(3,8)=0.0) LNG 461
    DATA(S=3.676E-8,4.158E-8,4.644E-8,5.051E-8,5.056E-8,3.546E-8, LNG 462
    15.389E-8,5.706E-8) LNG 463
    DATA(A=2.755E+5,7.773E+5,14.165E+5,22.733E+5,21.279E+5,1.718E+5, LNG 464
    130.550E+5,42.946E+5) LNG 465
    DATA(0=1.,1.5,1.67,1.83,1.79,1.03,1.91,2.11) LNG 466
C THESE ARE THE ACENTRICITY FACTORS (FOR MOLECULAR SHAPES, ETC.) *** LNG 467
    DATA(Y=35.,45.,60.,75.,75.,40.,90.,105.) LNG 468
    DATA(P1=3.14159),(N1=6.025E+23),(R=8.3143 ) LNG 469
    IF(KEY.EQ.0)GO TO 1 LNG 470
    KEY=0 LNG 471
    DO 2 I=1,8 LNG 472
    J(I,I)=0.0 LNG 473
    K(I,I)=0.0 LNG 474
    DO 2 M=I,8 LNG 475
    J(M,I)=J(I,M) LNG 476
    2 K(M,I)=K(I,M) LNG 477
    1 CONTINUE LNG 478
    P=Q*.1 LNG 479
    IW=8 LNG 480
    DO 10 I=1,IW LNG 481
    10 E(I)=0(I) LNG 482
    DO 15 I=1,IW LNG 483
    15 B(I)=(2./3.)*P1*N1*S(I)**3 LNG 484
    DO 20 I=1,IW LNG 485
    DO 20 M=1,IW LNG 486
    D(I,M) =((B(I)**( 1./3. ) + B(M) ** ( 1./3.))/ 2. )*(1.-J(I,M) )) LNG 487
    1 **3 LNG 488
    20 C( I , M ) = ( 1. - K( I , M ) ) * ( A(I)* A(M) ) ** ( 1. / 2. ) * LNG 489
    1( D( I , M ) **2 / ( B(I) * B(M) ) ) ** ( 1./2. ) LNG 490
    A2=0 $ B2=0 LNG 491
    E2=0 $ V2=0 LNG 492
    DO 25 I=1,IW LNG 493
    V2=V2+X(I)*Y(I) LNG 494
    DO 25 M=1,IW LNG 495
    E2=E2+X(I)*X(M)*(E(I)+E(M))/2. LNG 496
    A2 = A2 + X(I) * X(M) * C( I , M ) LNG 497
    25 B2=B2+X(I)*X(M)*D(I,M) LNG 498
    V6=V2 LNG 499
    DO 30 I = 1, IW LNG 500
    A3 = A(I) LNG 501
    B3 = B(I) LNG 502
    E3 = E(I) LNG 503
    V2 = Y(I) LNG 504

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V1 = FIND V1( A3, B3, E3, R, V2, P, T) LNG 505
G1 = FIND G1( A3, B3, E3, R, V2, P, T) LNG 506
V(I) = V1 LNG 507
30 G(I) = G1 LNG 508
A3=A2 LNG 509
B3=B2 LNG 510
E3=E2 LNG 511
V2=V6 LNG 512
V1=FIND V1(A3,B3,E3,R,V2,P,T) LNG 513
C G1=FIND G1(A3,B3,E3,R,V2,P,T) LNG 514
V7=V1 LNG 515
G7=G1 LNG 516
C EXCEAS VOLUME AND GIBS ENERGY LNG 517
V9=0 LNG 518
G9=0 LNG 519
DO 35 I=1,IW LNG 520
V9=V9-X(I)*V(I) LNG 521
35 G9=G9-X(I)*G(I) LNG 522
W9 = -V9 LNG 523
H9 = -G9 LNG 524
V9=V9+V7 LNG 525
G9=G9+G7 LNG 526
VIWW=V(IWW) LNG 527
RETURN LNG 528
END LNG 529
FUNCTION FIND V1( A3, B3, E3,R, V2, P, T ) LNG 530
C SOLVES THE HARD SPHERE EQUATION OF STATE FOR VOLUME GIVEN P AND T LNG 531
C A2 IS THE CONSTANT A, B2 IS THE CONSTANT B, E3 IS THE ACENTRICITY LNG 532
INDEX = 0 LNG 533
1 V1 = V2 LNG 534
X1 = B3 / ( 4. * V1 ) LNG 535
F2 = (( 1. + X1 + X1**2) / ( 1. - X1)**3) * E3 -A3 / ( V1 * R * T ) LNG 536
1- (P * V1) / ( R*T ) LNG 537
F3 = A3 / (R*T*V1**2) - P/ (R*T) LNG 538
F3 = F3 - ((( X1 + 2.* X1**2) * ( 1 - X1 ) + 3* ( 1. + X1 + X1**2 )
1* X1) / ( ( 1. - X1 )**4 * V1)) * E3 LNG 539
V2 = V1 - F2/ F3 LNG 540
IF( ABS( (V2 - V1) / V2) .LT. .00001 ) GO TO 2 LNG 541
INDEX = INDEX + 1 LNG 542
IF( INDEX .LT. 250 ) GO TO 1 LNG 543
2 V1 = V2 LNG 544
FIND V1 = V2 LNG 545
RETURN LNG 546
END LNG 547
FUNCTION FIND G1( A3, B3, E3,R, V2, P, T ) LNG 548
C CALCULATES THE GIBBS FREE ENERGY FOR THE HARD SPHERE EOS LNG 549
V1 = V2 LNG 550
X1 = B3 / ( 4. * V1 ) LNG 551
G1 =ALOG( 1. / ( 1. - X1 ) ) +(3. *X1) / ( 1. - X1 ) + ( 3. * X1**2 ) LNG 552
1 / (2. * ( 1. - X1 )**2) LNG 553
G1 = G1 - A3 / ( E3 * R * T*V1 ) + ( P * V1) / ( E3 * R * T ) -1.0 LNG 554
1 - ALOG( V1 ) LNG 555
G1 = R * T * E3 * G1 LNG 556
FIND G1 = G1 LNG 557
RETURN LNG 558
END LNG 559
FUNCTION EXCESS(P,DD,T,X) LNG 560
C CALCULATES THE EXCESS OR IDEAL VOLUM DEPENDING ON THE ENTRY LNG 561
DIMENSION X(10),F(10) LNG 562
KR=0 LNG 563
GO TO 1 LNG 564
ENTRY V IDEL LNG 565
KR=1 LNG 566

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1 CONTINUE LNG 568
CALL ZERO(F) LNG 569
IF(X(1).LE..000001)GO TO 2 LNG 570
CALL DATA CH4 LNG 571
IF(T.GT.190.555)GO TO 12 LNG 572
PM=VPN(T)+.00001 LNG 573
DELP=P-PM LNG 574
D=SAT(T,1) LNG 575
CALL DPDD(DP,D,T) LNG 576
DELD=DELP/DP LNG 577
D=DELD+D LNG 578
F(1)=X(1)*1000./D LNG 579
GO TO 2 LNG 580
12 D=FIND M(P,T,DD) LNG 581
IF(D.LE.0.0)D=1000. LNG 582
F(1)=X(1)*1000./D LNG 583
2 IF(X(2).LE. .000001)GO TO 3 LNG 584
F(2)=X(2)*1000./SAT(T,2) LNG 585
3 IF(X(3).LE..000001)GO TO 4 LNG 586
F(3)=X(3)*1000./SAT(T,3) LNG 587
4 IF(X(4).LE..00001)GO TO 5 LNG 588
F(4)=X(4)*1000./SAT(T,4) LNG 589
5 IF(X(5).LE..00001)GO TO 6 LNG 590
F(5)=X(5)*1000./SAT(T,5) LNG 591
6 IF(X(7).LE..00001)GO TO 61 LNG 592
F(7)=X(7)*1000./SAT(T,7) LNG 593
61 IF(X(8).LE..00001)GO TO 62 LNG 594
F(8)=X(8)*1000./SAT(T,8) LNG 595
62 IF(X(6).LE..00001)GO TO 8 LNG 596
CALL DATA N2 LNG 597
IF(T.GT.126.6)GO TO 7 LNG 598
PN=VPN(T)+.00001 LNG 599
DELP=P-PN LNG 600
D=SAT(T,6) LNG 601
CALL DPDD(DP,D,T) LNG 602
F(6)=X(6)*1000./(D+DELP/DP) LNG 603
GO TO 8 LNG 604
7 D=FIND M(P,T,DD) LNG 605
IF(D.LE.0.0)D=1000. LNG 606
F(6)=X(6)*1000./D LNG 607
8 V=1000./DD LNG 608
VS=0 LNG 609
DO 21 I=1,8 LNG 610
21 VS=VS+F(I) LNG 611
EXCESS=V-VS LNG 612
IF(KR.GT.0)EXCESS=VS LNG 613
CALL DATA CH4 LNG 614
RETURN LNG 615
END LNG 616
SUBROUTINE ZERO(X) LNG 617
C INITIALIZES THE COMPONENT MATRIX TO 0 LNG 618
DIMENSION X(10) LNG 619
DO 1 I=1,10 LNG 620
1 X(I)=0.0 LNG 621
RETURN LNG 622
END LNG 623
FUNCTION FMKM(T,X) LNG 624
C THE REVISED KLOSEK AND MCKINLEY METHOD,THE INPUT IS TEMPERATURE LNG 625
C AND THE COMPONENT MATRIX. TEMPERATURE IS IN KELVIN,OUTPUT IS LNG 626
C DENSITY IN MOLES PER LITER. THE ALLOWABLE COMPONENTS ARE C1,C2,C3 LNG 627
C NC4,IC4,N2,NC5,IC5 IN THAT ORDER. THIS METHOD SHOULD NOT BE USED LNG 628
C FOR MIXTURES WITH LESS THAN 60% METHANE, OR FOR MIXTURES CONTAINING LNG 629
C MORE THAN 4% NITROGEN OR MORE THAN 4% EACH OF NC4 OR IC4 OR MORE LNG 630

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C THAN 2% TOTAL OF NC5 AND IC5. LNG 631
DIMENSION TM(100),TN(100),X(10),Q(8) LNG 632
DATA(0=16.04303,30.07012,44.09721,58.1243,58.1243,28.0134,72.1513 LNG 633
19,72.15139) LNG 634
DATA((TM(I),I=1,10)=-.005,.12,.22,.34,.43,.515,.595,.66,.725,.795) LNG 635
DATA((TM(I),I=11,20)=-.006,.135,.26,.38,.5,.59,.665,.74,.81,.885) LNG 636
DATA((TM(I),I=21,30)=-.007,.15,.3,.425,.575,.675,.755,.83,.91,.99) LNG 637
DATA((TM(I),I=31,40)=-.007,.165,.34,.475,.635,.735,.84,.92,1.045, A1.12) LNG 638
A1.245) LNG 639
DATA((TM(I),I=41,50)=-.008,.19,.375,.535,.725,.835,.95,1.055,1.155 LNG 640
A1.38) LNG 641
DATA((TM(I),I=51,60)=-.009,.22,.44,.61,.81,.945,1.065,1.18,1.28, A1.55) LNG 642
A1.86) LNG 643
DATA((TM(I),I=61,70)=-.01,.25,.5,.695,.92,1.055,1.205,1.33,1.45, A1.55) LNG 644
A1.71) LNG 645
DATA((TM(I),I=71,80)=-.013,.295,.59,.795,1.035,1.21,1.385,1.525, 11.64,1.75) LNG 646
DATA((TM(I),I=81,90)=-.015,.345,.7,.92,1.2,1.37,1.555,1.715, A1.86,1.99) LNG 647
DATA((TM(I),I=91,100)=-.017,.4,.825,1.06,1.39,1.59,1.8,1.95, A2.105,2.272) LNG 648
A2.12) LNG 649
DATA((TN(I),I=1,10)=-.004,.1,.22,.35,.5,.6,.69,.78,.86,.95) LNG 650
DATA((TN(I),I=11,20)=-.005,.12,.28,.43,.59,.71,.83,.94,1.05,1.14) LNG 651
DATA((TN(I),I=21,30)=-.007,.16,.34,.49,.64,.79,.94,1.08,1.17,1.27) LNG 652
DATA((TN(I),I=31,40)=-.01,.24,.42,.61,.75,.91,1.05,1.19,1.33,1.45) LNG 653
DATA((TN(I),I=41,50)=-.015,.32,.59,.77,.92,1.07,1.22,1.37,1.52, 11.71) LNG 654
DATA((TN(I),I=51,60)=-.024,.41,.72,.95,1.15,1.22,1.3,1.45,1.65,2.) LNG 655
DATA((TN(I),I=61,70)=-.032,.6,.91,1.23,1.43,1.63,1.85,2.08,2.3, 12.45) LNG 656
DATA((TN(I),I=71,80)=-.043,.71,1.13,1.48,1.73,1.98,2.23,2.48,2.75 1,2.9) LNG 657
DATA((TN(I),I=81,90)=-.058,.95,1.46,1.92,2.2,2.42,2.68,3., A3.32,3.52) LNG 658
DATA((TN(I),I=91,100)=-.075,1.3,2.,2.4,2.6,3.,3.4,3.77, A3.99,4.23) LNG 659
IF(X(1).LT..00001)GO TO 20 LNG 660
AW=0.0 LNG 661
DO 1 I=1,8 LNG 662
1 AW=AW+X(I)*Q(I) LNG 663
VI=VIDEAL(T,X) LNG 664
J=1 LNG 665
IF(T.GE.95.)J=11 LNG 666
IF(T.GE.100.)J=21 LNG 667
IF(T.GE.105.)J=31 LNG 668
IF(T.GE.110.)J=41 LNG 669
IF(T.GE.115.)J=51 LNG 670
IF(T.GE.120.)J=61 LNG 671
IF(T.GE.125.)J=71 LNG 672
IF(T.GE.130.)J=81 JJ=J+9 LNG 673
JJ=J+9 LNG 674
W=15. DO 5 I=J,JJ LNG 675
W=W+1. IF(AW.GT.W)GO TO 5 LNG 676
GO TO 6 LNG 677
5 CONTINUE I=JJ LNG 678
I=JJ LNG 679
6 DIF1=AW-W J=I-1 LNG 680
FK=(TM(I)-TM(J))*DIF1+TM(I) LNG 681
FK1=(TM(I+10)-TM(J+10))*DIF1+TM(I+10) LNG 682
IT=(T+.00001)/5. LNG 683

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DIF2=T-IT*5 LNG 694
IF(T.GE.135.)DIF2=T-130. LNG 695
IF(T.LT.90.)DIF2=T-90. LNG 696
FK=FK+(FK1-FK)*DIF2/5. LNG 697
IF(X(6).LT..0001)GO TO 17 LNG 698
FKN=(TN(I)-TN(J))*DIF1+TN(I) LNG 699
FK1=(TN(I+10)-TN(J+10))*DIF1+TN(I+10) LNG 700
FKN=FKN+(FK1-FKN)*DIF2/5. LNG 701
FK=FK+(FKN-FK)*X(6)/.0425 LNG 702
17 FK=FK/1000. LNG 703
FMKM=1./(VI-FK*X(1)) LNG 704
RETURN LNG 705
20 FMKM=0.0 LNG 706
RETURN LNG 707
END LNG 708
FUNCTION V IDEAL(T,X) LNG 709
C CALCULATES THE IDEAL VOLUME OF A MIXTURE FOR THE K AND M METHOD LNG 710
DIMENSION X(10) LNG 711
V=0 LNG 712
J=0 LNG 713
IF(X(6).GT..0001.AND.T.GT.115.)J=1 LNG 714
DO 10 I=1,8 LNG 715
IF(X(I).LE..000001)GO TO 10 LNG 716
IF(J.GT.0.AND.I.EQ.6)GO TO 10 LNG 717
V=V+X(I)/SAT(T,I) LNG 718
10 CONTINUE LNG 719
IF(J.EQ.1)V=V+X(6)/SATN2(T) LNG 720
VIDEAL=V LNG 721
RETURN LNG 722
END LNG 723
FUNCTION SATN2(T) LNG 724
C CALCULATES A PSEUDO SATURATED LIQUID DENSITY FOR N2 ABOVE 115 K LNG 725
IF(T.LT.115.)GO TO 1 LNG 726
DELT=(T-115.) LNG 727
SATN2=SAT(115.,6)+DELT*(SAT(115.05,6)-SAT(114.95,6))/1 LNG 728
RETURN LNG 729
1 SATN2=SAT(T,6) LNG 730
RETURN LNG 731
END LNG 732
FUNCTION SAT(T,I) LNG 733
C CALCUALTES THE PURE FLUID DENSITIES FOR THE K AND M METHOD LNG 734
UNITS ARE DEG K AND MOLES/LITER LNG 735
DIMENSION A(7,8) LNG 736
DATA((A(I),I=1,7)=190.555,10.16,18.65812322,6.712030737, LNG 737
1,-.9472019702,0.0,0.0) LNG 738
DATA((A(I),I=8,14)=305.33, LNG 739
1 6.86,12.55205121,13.43284373,-19.00461066,11.07715985,0.0) LNG 740
DATA((A(I),I=15,21)=369.82,5.,8.684458671,18.04085714,-29.46261356 LNG 741
1,16.43559114,0.0) LNG 742
DATA((A(I),I=22,28)=425.16,3.92, LNG 743
1 7.286062567,11.96307859,-19.87591962, LNG 744
211.60211932,0.0) LNG 745
DATA((A(I),I=29,35)=408.13,3.8, LNG 746
1 7.657535400,8.145251283,-13.10582462, LNG 747
28.145894091,0.0) LNG 748
DATA((A(I),I=36,42)=126.2,11.21,19.39216835,26.01408462,-39.497587 LNG 749
191.23.32977312,0.0) LNG 750
DATA((A(I),I=43,49)=469.6,3.285,-.0362004993,59.00202990, LNG 751
1-93.44193819,43.66780833,0.0) LNG 752
DATA((A(I),I=50,56)=460.39,3.271,2.946310456,35.50770979, LNG 753
1-57.41242993,28.15898339,0.0) LNG 754
IF(T.GT.A(1,I))GO TO 1 LNG 755
X=(1.-T/A(1,I)) LNG 756

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SAT=(A(2,I)+A(3,I)*X**(.35)+A(4,I)*X+A(5,I)*X**(.3/.3)
1+A(6,I)*X**(.3/.3))
RETURN
1 SAT=1.E+20
RETURN
END
SUBROUTINE ECKNON(PIN,DOUT,TIN,Q)
MAIN: ECKERT-RENON DENSITY MODEL
C
C
C THIS PROGRAM IS DESIGNED FOR THE CALCULATION OF
C LNG LIQUID DENSITIES ONLY AND WILL NOT
C COMPUTE DENSITIES ABOVE 160 K.
C
C THIS PROGRAM IS CALIBRATED SOLELY AGAINST DATA
C MEASURED BY
C UNITED STATES DEPARTMENT OF COMMERCE
C NATIONAL BUREAU OF STANDARDS
C CRYOGENICS LABORATORY
C BOULDER, COLORADO
C
C
C >>>>>>>>>NOTICE<<<<<<<<<<<<<<<<<<<<<
C NO WARRANTY IS MADE OR IMPLIED AS TO THE ACCURACY
C OF DENSITIES CALCULATED BY USE OF THIS MODEL.
C
C USERS ARE AT THEIR OWN RISK IN THE USE OF THIS PROGRAM.
C
C
C COMPONENT ID NUMBERS ARE
C 1 METHANE
C 2 ETHANE
C 3 PROPANE
C 4 N-BUTANE
C 5 I-BUTANE
C 6 NITROGEN
C 7 N-PENTANE
C 8 I-PENTANE
C
C
C INPUT VARIABLES ARE
C PIN=PRESSURE IN BAR
C TIN=TEMPERATURE IN K
C Q=MATRIX OF MOLE FRACTIONS OF COMPONENTS
C OUTPUT VARIABLE IS
C DOUT=DENSITY IN MOLES/LITER
C DIMENSION VHOLD(12),VPHOLD(12),Q(8)
C TYPE INTEGER CNT1,CNT2,SWITCH
C TYPE REAL MWM,INCR,KIJ,MW,MOL,LAM,LAMB
C COMMON /RUN/ID(12),X(12),NAM(2,12),C12(12,12),R(12),Z(12),S(12),VS
1(12),TS(12),US(12),SG(12),EP(12),LAM(12),NC,MW(12),AIJ(12,12),PST(
212),TCT(12)
C COMMON /DAT/TBI(12),VBI(12),TBIM(12),VBIM(12),C(12),SGIM(12),SIM(1
12),EPIM(12),PIM(12),SIN(12),VSIM(12),USIM(12),TSIM(12),DVBIM(12),P
2D(12),PV(12),RHO(12),VBIMP(12),TTM(12),MWM,CNT1,CNT2,TOLD(12),INCR
3(12),SWITCH,JPC,JPCS,JMIX,DENS,VEX,HEX,GEX,TMP,VMP,TP
C COMMON/PAR/KIJ( 8, 8),AJI( 8, 8)

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COMMON/UNITS/ITC,IPC,T,TMAX,DT,P,PMAX,DP,PS          LNG 820
SWITCH = 0                                         LNG 821
1 CALL ZERO1                                     LNG 822
C THE NEXT VALUES ARE SET TO PREVENT UNDER/OVER FLOWS LNG 823
TMAX = 100.0                                      LNG 824
DT = 100.0                                       LNG 825
PMAX = 100.0                                      LNG 826
DP = 100.0                                         LNG 827
PS = 100.0                                         LNG 828
CALL INPUT(Q)                                     LNG 829
CODE=0                                            LNG 830
ITC=0                                             LNG 831
IPC=0                                             LNG 832
P=PIN/1.01325                                    LNG 833
T=TIN                                            LNG 834
IF(T.GT.(160.)) GO TO 19                         LNG 835
8 CALL PZERO(CTMX)                                LNG 837
MWM = 0                                           LNG 838
DO 9 I = 1,NC                                     LNG 839
MWM = MWM + (X(I)*MW(I))                         LNG 840
9 CONTINUE                                         LNG 841
10 DO 11 I = 1,NC                                 LNG 842
TBI(I) = T / TS(I)                               LNG 843
CALL VOLUME(TBI(I),VBI(I),JPUR,NC)               LNG 844
TBIM(I) = T / TSIM(I)                            LNG 845
CALL VOLUME(TBIM(I),VBIM(I),JMIX,NC)             LNG 846
11 CONTINUE                                         LNG 847
VM = 0.0                                           LNG 848
DO 12 I = 1,NC                                     LNG 849
VM = VM + (X(I)*VBIM(I)*VSIM(I))                LNG 850
12 CONTINUE                                         LNG 851
TMP = ( T * 1.8 ) - (459.67)                      LNG 852
DO 13 I = 1,NC                                     LNG 853
VHOLD(I) = VBI(I)                                LNG 854
VPHOLD(I) = VBIM(I)                             LNG 855
PP = (P*TS(I))/US(I)                            LNG 856
VBIMP(I) = VBI(I)                                LNG 857
CALL PRES(PP,VBIMP(I),TBI(I),NC)                 LNG 858
VBI(I) = VBIMP(I)                                LNG 859
PP = (P * TSIM(I)) / USIM(I)                     LNG 860
VBIMP(I) = VBIM(I)                                LNG 861
CALL PRES(PP,VBIMP(I),TBIM(I),NC)                 LNG 862
VBIM(I) = VBIMP(I)                                LNG 863
13 CONTINUE                                         LNG 864
VMP = 0.0                                           LNG 865
DO 14 I = 1,NC                                     LNG 866
VMP = VMP + (X(I)*VBIMP(I)*VSIM(I))              LNG 867
PD(I) = VBIMP(I) * VSIM(I)                        LNG 868
14 CONTINUE                                         LNG 869
DENS=MWM/VMP                                      LNG 870
DOUT=DENS*1000./MWM                                LNG 871
RETURN                                              LNG 872
19 DOUT=0.0                                         LNG 873
RETURN                                              LNG 874
END                                                 LNG 875
SUBROUTINE ZERO1                                  LNG 876
COMMON /RUN/A(361)                                LNG 877
COMMON /DAT/B(224)                                LNG 878
DO 1 I = 1,361                                     LNG 879
1 A(I) = 0.0                                       LNG 880
DO 2 I = 1,224                                     LNG 881
2 B(I) = 0.0                                       LNG 882

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RETURN                                              LNG 883
END                                                 LNG 884
SUBROUTINE INPUT(Q)                               LNG 885
DIMENSION NNO( 8),SIGM( 8),EPSI( 8),LAMB( 8),NAME(2, 8),SNO( 8),MO
1L( 8),CT( 8),TCH( 8),VCH( 8),ECH( 8),AR( 8),AZ( 8),Q(8)      LNG 886
1L( 8),CT( 8),TCH( 8),VCH( 8),ECH( 8),AR( 8),AZ( 8),Q(8)      LNG 887
TYPE INTEGER CNT1,CNT2,SWITCH                      LNG 888
TYPE REAL MWM,INCR,KIJ,MW,MOL,LAM,LAMB            LNG 889
COMMON /RUN/ID(12),X(12),NAM(2,12),C12(12,12),R(12),Z(12),VS
1(12),TS(12),US(12),SG(12),EP(12),LAM(12),NC,MW(12),AIJ(12,12),PST(
212),TCT(12)                                         LNG 890
COMMON /DAT/TBI(12),VBI(12),TBIM(12),VBIM(12),C(12),SGIM(12),SIM(1
12),EPIM(12),PIM(12),SIN(12),VSIM(12),USIM(12),TSIM(12),DVBIM(12),P
2D(12),PV(12),RHO(12),VBIMP(12),TTM(12),MWM,CNT1,CNT2,TOLD(12),INCR
3(12),SWITCH,JPC,JPCS,JMIX,DENS,VEX,HEX,GEX,TMP,VMP,TP          LNG 891
COMMON/PAR/KIJ( 8, 8),AJI( 8, 8)                  LNG 892
DATA SIGM/                                         LNG 893
*0.991000,1.029000,1.155000,1.278000,1.388752,1.392995,          LNG 894
*1.47162,1.47217/                                LNG 895
DATA EPSI/                                         LNG 896
*0.640000,0.909000,1.69800,2.237000,2.705262,2.545907,          LNG 897
*3.76195,3.68336/                                LNG 898
DATA LAMB/                                         LNG 899
*1.053604,0.986325,1.227117,1.408519,1.473346,1.461676,          LNG 900
*1.698790,1.695012/                                LNG 901
DATA MOL/                                         LNG 902
*28.01600,16.04200,30.06800,44.09400,58.12000,58.12000,          LNG 903
*72.146000,72.146000/                                LNG 904
DATA SNO/                                         LNG 905
*10.00000,10.00000,10.00000,10.00000,10.00000,10.00000,          LNG 906
*10.00000,10.00000/                                LNG 907
DATA CT/                                           LNG 908
*126.0600,190.5600,305.4300,369.8200,425.1600,408.0300,          LNG 909
*469.65,460.39/                                LNG 910
DATA TCH/                                         LNG 911
*112.7699,170.9645,256.8311,294.7255,340.7867,322.6109,          LNG 912
*411.0100,403.3200/                                LNG 913
DATA VCH/                                         LNG 914
*26.01846,29.04010,40.88188,54.60301,68.72075,69.44272,          LNG 915
*84.90197,84.99830/                                LNG 916
DATA ECH/                                         LNG 917
*1393.000,1977.000,3695.000,4867.000,5886.652,5539.895,          LNG 918
*8186.000,8015.000/                                LNG 919
DATA AR/                                           LNG 920
*1.000000,1.000000,1.000000,1.000000,1.000000,1.000000,          LNG 921
*1.000000,1.000000/                                LNG 922
DATA AZ/                                           LNG 923
*10.00000,10.00000,10.00000,10.00000,10.00000,10.00000,          LNG 924
*10.00000,10.00000/                                LNG 925
DO 4 I=2,6                                         LNG 926
4 X(I)=Q(I-1)                                     LNG 927
X(1)=Q(6)                                         LNG 928
X(7)=Q(7)                                         LNG 929
X(8)=Q(8)                                         LNG 930
NC=0                                              LNG 931
DO 1 I=1,8                                         LNG 932
IF(X(I).LE.0.0)GO TO 1                           LNG 933
NC=NC+1                                         LNG 934
X(NC)=X(I)                                         LNG 935
ID(NC)=I                                         LNG 936
1 CONTINUE                                         LNG 937
DO 2 I = 1,NC                                      LNG 938
J = ID(I)                                         LNG 939
R(I) = AR(J)                                       LNG 940
                                             LNG 941
                                             LNG 942
                                             LNG 943
                                             LNG 944
                                             LNG 945

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Z(I) = AZ(J)	LNG 946
S(I) = SNO(J)	LNG 947
TS(I)= TCH(J)	LNG 948
US(I)= ECH(J)	LNG 949
SG(I) = SIGM(J)	LNG 950
TCT(I) = CT(J)	LNG 951
EP(I) = EPSI(J)	LNG 952
LAM(I) = LAMB(J)	LNG 953
VS(I) = VCH(J)	LNG 954
MW(I) = MOL(J)	LNG 955
2 CONTINUE	LNG 956
DO 3 I = 1,NC	LNG 957
DO 3 J = 1,NC	LNG 958
M = ID(I)	LNG 959
L = ID(J)	LNG 960
C12(I,J) = KIJ(M,L)	LNG 961
C12(J,I) = KIJ(L,M)	LNG 962
AIJ(I,J) = AJI(M,L)	LNG 963
AIJ(J,I) = AJI(L,M)	LNG 964
3 CONTINUE	LNG 965
RETURN	LNG 966
END	LNG 967
END	LNG 968
SUBROUTINE PZERO(TRT)	LNG 969
TYPE INTEGER CNT1,CNT2,SWITCH	LNG 970
TYPE REAL MWM,INCR,KIJ,MW,MOL,LAM,LAMB	LNG 971
COMMON /RUN/ID(12),X(12),NAM(2,12),C12(12,12),R(12),Z(12),S(12),VS	LNG 972
1(12),TS(12),US(12),SG(12),EP(12),LAM(12),NC,MW(12),AIJ(12,12),PST(LNG 973
212),TCT(12)	LNG 974
COMMON /DAT/TBI(12),VBI(12),TBIM(12),VBIM(12),C(12),SGIM(12),SIM(1	LNG 975
12),EPIM(12),PIM(12),SIN(12),VSIM(12),USIM(12),TSIM(12),DVBIM(12),P	LNG 976
2D(12),PV(12),RHO(12),VBIMP(12),TTM(12),MWM,CNT1,CNT2,TOLD(12),INCR	LNG 977
3(12),SWITCH,JPC,JPCS,JMIX,DENS,VEX,HEX,GEX,TMP,VMP,TP	LNG 978
COMMON/PAR/KIJ(8, 8),AJI(8, 8)	LNG 979
COMMON /UNITS/ITC,IPC,T,TMAX,DT,P,PMAX,DP,PS	LNG 980
IF(NC.EQ.1) TRT = T/TCT(1)	LNG 981
IF(NC.GT.1) CALL TCM(TRT)	LNG 982
IF(NC.GT.1) TRT = T/TRT	LNG 983
IF(NC.EQ.1) CALL PURE(TRT,TRT,R(1),RXX)	LNG 984
DO 1 I = 1,NC	LNG 985
TCXX = T/TCT(I)	LNG 986
IF(NC.GT.1) CALL PURE(TCXX,TRT,R(I),RTR)	LNG 987
JPC = 0	LNG 988
IF(TRT.GT.(.87)) JPC = 2	LNG 989
IF(TRT.GT.(1.0)) JPC = 1	LNG 990
IF(NC.EQ.1) RTR = RXX	LNG 991
C THERE ARE SEVERAL OTHER FORMS OF THESE EQUATIONS	LNG 992
C REFER TO ORIGINAL ARTICLES FOR VALUES TO USE	LNG 993
S(I) = RTR * Z(I) - 2. * RTR + 2.	LNG 994
SIM(I) = S(I)	LNG 995
SG(I) = (VS(I)/RTR) **(1./3.)	LNG 996
EP(I) = US(I) / S(I)	LNG 997
C(I) = LAM(I) * (2176./((185.6)*(1.98726)))	LNG 998
TTM(I) = TS(I)	LNG 999
TOLD(I) = TS(I)	LNG1000
INCR(I) = (20.0)	LNG1001
1 CONTINUE	LNG1002
JPUR = 0	LNG1003
CNT1 = 1	LNG1004
CNT2 = 1	LNG1005
JMIX = 0	LNG1006
2 DEN = 0	LNG1007
	LNG1008

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DO 3 I = 1,NC          LNG1009
DEN = DEN + (X(I) * SIM(I))    LNG1010
3 CONTINUE             LNG1011
DO 4 I = 1,NC          LNG1012
PIM(I) = (X(I) * SIM(I)) / DEN   LNG1013
4 CONTINUE             LNG1014
DO 6 I = 1,NC          LNG1015
SGIM(I) = 0.0           LNG1016
DO 6 J = 1,NC          LNG1017
M = I                  LNG1018
N = J                  LNG1019
IF(I.GT.J) M = J        LNG1020
IF(I.GT.J) N = I        LNG1021
IF(I.EQ.J) GO TO 5      LNG1022
TRUDX = (AIJ(N,M)/(T/TTM(I)))    LNG1023
IF(TRUDX.LT.(-180.0)) GO TO 5     LNG1024
SGIM(I) = SGIM(I) +((PIM(J) *(((SG(I)**(1./3.) + SG(J)**(1./3.))
1/(2.0))**3))* (AIJ(M,N) * EXP(AIJ(N,M)/(T/TTM(I)))))    LNG1025
1 GO TO 6               LNG1026
5 SGIM(I) = SGIM(I) +(PIM(J) *(((SG(I)**(1./3.) + SG(J)**(1./3.))
1/(2.0))**3 )          LNG1027
LNG1028
6 CONTINUE             LNG1029
DEN = 0.0               LNG1030
DO 7 I = 1,NC          LNG1031
DEN = DEN + (PIM(I)* SGIM(I)* SGIM(I))    LNG1032
7 CONTINUE             LNG1033
DO 8 I = 1,NC          LNG1034
SIN(I) =(S(I) * (SGIM(I)**(2.))) / DEN    LNG1035
8 CONTINUE             LNG1036
LNG1037
DO 9 I = 1,NC          LNG1038
TEST = (1.0) - (SIM(I) / SIN(I))    LNG1039
TEST = ABS (TEST)           LNG1040
IF(TEST.GT.(0.00001)) GO TO 10      LNG1041
9 CONTINUE             LNG1042
GO TO 12               LNG1043
10 DO 11 I = 1,NC         LNG1044
SIM(I) =(SIM(I) + SIN(I)) /(2.0)    LNG1045
LNG1046
11 CONTINUE             LNG1047
CNT1 = CNT1 + 1          LNG1048
IF(CNT1.GT.250) GO TO 12      LNG1049
GO TO 2                 LNG1050
12 DO 14 I = 1,NC         LNG1051
SIM(I) = SIN(I)           LNG1052
EPIM(I) = 0.0              LNG1053
DO 14 J = 1,NC          LNG1054
K = I                  LNG1055
L = J                  LNG1056
IF(J.LT.I) K = J          LNG1057
IF(J.LT.I) L = I          LNG1058
IF(I.EQ.J) GO TO 13      LNG1059
TRUDX = C12(L,K) / (T/TTM(I))    LNG1060
IF(TRUDX.LT.(-180.0)) GO TO 13      LNG1061
EPIM(I) = EPIM(I) +((PIM(J) * SQRT(EP(I)*EP(J)))*
1(C12(K,L) * EXP((C12(L,K) /(T/TTM(I))))))    LNG1062
1 GO TO 14               LNG1063
13 EPIM(I) = EPIM(I) + (PIM(J) * SQRT(EP(I) * EP(J)))    LNG1064
LNG1065
14 CONTINUE             LNG1066
C ASSUME NUMBER OF MOLES OF MIXTURE = 1.0      LNG1067
DO 15 I = 1,NC          LNG1068
VSIM(I) = R(I) * (SGIM(I)**(3.0))    LNG1069
USIM(I) = SIM(I) * EPIM(I)           LNG1070
TSIM(I) = USIM(I) / ((1.98726) * C(I))    LNG1071

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15 CONTINUE LNG1072
   DO 16 I = 1,NC LNG1073
   TEST = (1.0) - (TSIM(I)/TTM(I)) LNG1074
   TEST = ABS(TEST) LNG1075
   IF(TEST.GT.(0.00001)) GO TO 17 LNG1076
16 CONTINUE LNG1077
   RETURN LNG1078
17 DO 20 I = 1,NC LNG1079
   IF(TSIM(I) - TTM(I))18,20,19 LNG1080
18 TNEW = TTM(I) - INCR(I) LNG1081
   IF(TNEW.EQ.TOLD(I)) INCR(I) = INCR(I)/ (2.0) LNG1082
   IF(TNEW.EQ.TOLD(I)) GO TO 18 LNG1083
   TOLD(I) = TTM(I) LNG1084
   TTM(I) = TNEW LNG1085
   GO TO 20 LNG1086
19 TNEW = TTM(I) + INCR(I) LNG1087
   IF(TNEW.EQ.TOLD(I)) INCR(I) = INCR(I) / (2.0) LNG1088
   IF(TNEW.EQ.TOLD(I)) GO TO 19 LNG1089
   TOLD(I) = TTM(I) LNG1090
   TTM(I) = TNEW LNG1091
20 CONTINUE LNG1092
   CNT2 = CNT2 + 1 LNG1093
   IF(CNT2.GT.250) RETURN LNG1094
   GO TO 2 LNG1095
END LNG1096
SUBROUTINE PURE(TZ,T,Z,R) LNG1097
DIMENSION A(15) LNG1098
IF(TZ.LT.(.8653525)) R = Z LNG1099
IF(TZ.LT.(.8653525)) GO TO 2 LNG1100
IF(T.LT.(.8653525)) R = Z LNG1101
IF(T.LT.(.8653525)) GO TO 2 LNG1102
IF(T.GT.(1.0)) TA = 1.0 LNG1103
IF(T.LE.(1.0)) TA = T LNG1104
DATA A/.9184780,-.1530647,-.1090050,.8073883,1.441803,
1-10.85944,-6.041687,51.26758,.9062500,-108.9805,28.88672,
2106.1406,-45.78125,-38.43750,20.56250/ LNG1105
R = 0.0 LNG1106
TT = (TA - (.9267292)) / (.7267517E-01) LNG1107
DO 1 K = 1,14 LNG1108
   R = R + A(16-K) LNG1109
   R = R * TT LNG1110
1 CONTINUE LNG1111
   R = R + A(1) LNG1112
2 RETURN LNG1113
END LNG1114
SUBROUTINE VOLUME(T,V,J,NC) LNG1115
TR = T LNG1116
IF(TR.GT.(1.00)) GO TO 4 LNG1117
V = 0.5 LNG1118
1 VT = (1.0) + (0.1*TR*V**4.0)) LNG1119
   ERR = (1.0) - (V/VT) LNG1120
   TEST = ABS(ERR) LNG1121
   IF(TEST.LE.(0.00001)) GO TO 2 LNG1122
   V = VT LNG1123
   GO TO 1 LNG1124
2 V = V**(3.0) LNG1125
3 RETURN LNG1126
4 V = (((((10.06600 *TR)-24.79837)*TR)+23.260722)*TR)-6.686880) LNG1127
   GO TO 3 LNG1128
END LNG1129
SUBROUTINE BETA(P,DV,T,V,KK) LNG1130

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TYPE REAL AF,BF,CF,DF,EF,AG,BG,CG,DG,P,DV,T,V,F,F1,G,G1,DEN,EG      LNG1135
TYPE REAL C,B1,B2,B3                                              LNG1136
DIMENSION C(11)                                              LNG1137
DATA AF/0.11566564E+02/,BF/-0.53510144E+01/,CF/-0.74598207E-01/,      LNG1138
1DF/0.67068653E+00/,EF/-0.11939887E+00/,AG/0.62721813E+01/,      LNG1139
2BG/0.47698365E+00/,CG/-0.16023080E+01/,DG/0.49837746E+00/,      LNG1140
3EG/-0.42639183E-01/                                              LNG1141
F = AF +(BF*V)+(CF*V*V)+(DF*V*V*V)+(EF*V*V*V*V)                      LNG1142
IF(F.LT.(0.0)) GO TO 1                                              LNG1143
F1 = (BF+(CF*V*2.E+00)+(DF*V*V*3.E+00)+(EF*V*V*V*4.E+00))*      LNG1144
1T * EXP(F)                                              LNG1145
G = AG+(BG*V)+(CG*V*V)+(DG*V*V*V)+(EG*V*V*V*V)                      LNG1146
IF(G.LT.(0.00)) GO TO 1                                              LNG1147
G1 = (BG+(CG*V*2.E+00)+(DG*V*V*3.E+00)+(EG*V*V*V*4.E+00)) * EXP(G      LNG1148
1)                                              LNG1149
G1 = -G1                                              LNG1150
DEN = F1 + G1                                              LNG1151
DV = (1.E+00) / DEN                                              LNG1152
IF(DV.GT.(0.0).OR.DV.LT.(-1.0E+00)) GO TO 1                          LNG1153
GO TO 2                                              LNG1154
DATA C/-1.18659822E+02,-.5509099E+00,.4172102E+01,-.8996686E+00,      LNG1155
1-.1500376E+01,-.1958324E+00,-.2788988E+01,.5195283E+00,      LNG1156
2.3734878E+01,.1361904E+00,.4671948E-01/                                              LNG1157
1 B1 = C(1) + C(2)*T + C(3)*V                                              LNG1158
B2 = C(4)*((V+C(5))**2)*EXP((C(6))*((T+C(7))**2))                      LNG1159
B3 = C(11) * (V+C(8))**C(9)/(T**C(10))                                              LNG1160
DV = B1 + B2 + B3                                              LNG1161
DV = EXP(DV)                                              LNG1162
DV = - DV                                              LNG1163
2 RETURN                                              LNG1164
END                                              LNG1165
SUBROUTINE PRES(A,B,C,NC)                                              LNG1166
TYPE REAL H,P,V,T,K1,K2,K3,K4,VT,PT                                              LNG1167
P = A                                              LNG1168
V = B                                              LNG1169
T = C                                              LNG1170
H = 1.E-02                                              LNG1171
IF(H.GT.P) H = P                                              LNG1172
ASSIGN 3 TO KK                                              LNG1173
1 CALL BETA(P,K1,T,V,NC)                                              LNG1174
K1 = H * K1                                              LNG1175
VT = V + ((0.5E+00) *K1)                                              LNG1176
PT = P + ((0.5E+00)*H)                                              LNG1177
CALL BETA(PT,K2,T,VT,NC)                                              LNG1178
K2 = K2 * H                                              LNG1179
VT = V + ((0.5E+00) *K2)                                              LNG1180
CALL BETA(PT,K3,T,VT,NC)                                              LNG1181
K3 = K3 * H                                              LNG1182
PT = P + H                                              LNG1183
VT = V + ( K3)                                              LNG1184
CALL BETA(PT,K4,T,VT,NC)                                              LNG1185
K4 = K4 * H                                              LNG1186
V = V + (((K1 + (2.E+00*K2) + (2.E+00*K3) + K4) / (6.E+00))) )      LNG1187
P = P - H                                              LNG1188
IF(P.EQ.(0.0)) GO TO 4                                              LNG1189
IF(P.LT.(0.0)) GO TO 2                                              LNG1190
GO TO 1                                              LNG1191
2 GO TO KK , (3,4)                                              LNG1192
3 P = P + H                                              LNG1193
H = P                                              LNG1194
ASSIGN 4 TO KK                                              LNG1195
GO TO 1                                              LNG1196
4 A = P                                              LNG1197

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B = V                         LNG1198
C = T                         LNG1199
RETURN                         LNG1200
END                           LNG1201
SUBROUTINE TCM(TMC)
TYPE REAL NUM                  LNG1202
TYPE INTEGER CNT1,CNT2,SWITCH   LNG1203
TYPE REAL MWM,INCR,KIJ,MW,MOL,LAM,LAMB   LNG1204
DIMENSION VCI(6),TH(12)          LNG1205
COMMON /RUN/ID(12),X(12),NAM(2,12),C12(12,12),R(12),Z(12),S(12),VS LNG1207
1(12),TS(12),US(12),SG(12),EP(12),LAM(12),NC,MW(12),AIJ(12,12),PST( LNG1208
212),TCT(12)                   LNG1209
COMMON /DAT/TBI(12),VBI(12),TBIM(12),VBIM(12),C(12),SGIM(12),SIM(1 LNG1210
12),EPIM(12),PIM(12),SIN(12),VSIM(12),USIM(12),TSIM(12),DVBIM(12),P LNG1211
2D(12),PV(12),RHO(12),VBIMP(12),TTM(12),MWM,CNT1,CNT2,TOLD(12),INCR LNG1212
3(12),SWITCH,JPC,JPCS,JMIX,DENS,VEX,HEX,GEX,TMP,VMP,TP                 LNG1213
COMMON/PAR/KIJ( 8, 8),AJI( 8, 8)           LNG1214
COMMON/UNITS/ITC,IPC,T,TMAX,DT,P,PMAX,DP,PS      LNG1215
DATA VCI/ 1.44,1.59,2.27,3.18,4.03,4.21/        LNG1216
DO 1 I = 1,NC                      LNG1217
J = ID(I)                         LNG1218
V = VCI(J)**(2./3.)                LNG1219
TH(I) = X(I) * V                  LNG1220
1 CONTINUE                         LNG1221
TTZ = 0.0                          LNG1222
DO 2 I = 1,NC                      LNG1223
TTZ = TTZ + TH(I)                 LNG1224
2 CONTINUE                         LNG1225
DO 3 I = 1,NC                      LNG1226
TH(I) = TH(I) / TTZ                LNG1227
3 CONTINUE                         LNG1228
SUM1 = 0.0                          LNG1229
SUM2 = 0.0                          LNG1230
K = NC - 1                         LNG1231
DO 5 I = 1,K                       LNG1232
L = I + 1                          LNG1233
DO 4 J = L,NC                      LNG1234
TTZ = (TCT(I)-TCT(J))/(TCT(I)+TCT(J))    LNG1235
TTZ= ABS(TTZ)                      LNG1236
T12 = (((((TTZ*(-3.038))+(5.443))*TTZ)+(-1.343))*TTZ)    LNG1237
1 + (0.287))*TTZ) - (.0076)         LNG1238
T12 = T12*(TCT(I) + TCT(J)) * (0.9)    LNG1239
SUM2 = SUM2 + ((2.)*TH(I)*TH(J)*T12)  LNG1240
4 CONTINUE                         LNG1241
SUM1 = SUM1 + (TH(I)*TCT(I)*(1.8))    LNG1242
5 CONTINUE                         LNG1243
TMC = SUM1 + SUM2 + (TH(NC)*TCT(NC)*(1.8)) LNG1244
V = 0.0                            LNG1245
DO 6 I = 1,NC                      LNG1246
J = ID(I)                         LNG1247
TH(I) = X(I) * VCI(J)              LNG1248
V = V + TH(I)                      LNG1249
6 CONTINUE                         LNG1250
DO 7 I = 1,NC                      LNG1251
TH(I) = TH(I) / V                  LNG1252
7 CONTINUE                         LNG1253
TMM = 0.0                          LNG1254
DO 9 I = 1,NC                      LNG1255
DO 8 J = 1,NC                      LNG1256
M = ID(I)                         LNG1257
N = ID(J)                         LNG1258
NUM = (VCI(M)**(1./3.)) * (VCI(N)**(1./3.)) LNG1259
                                         LNG1260

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NUM = SQRT(NUM) LNG1261
DEN = (0.5)*((VCI(M)**(1./3.))+(VCI(N)**(1./3.))) LNG1262
NUM = NUM / DEN LNG1263
NUM = NUM**(3.) LNG1264
AKIJ = (1.0) - NUM LNG1265
TCIJ = (1.0 - AKIJ) * SQRT(TCT(I)*TCT(J)*1.8*1.8) LNG1266
TMM = TMM + TH(I)*TH(J)*TCIJ LNG1267
8 CONTINUE LNG1268
9 CONTINUE LNG1269
TCMP = TMM + (10.0) LNG1270
ICT = 0 LNG1271
TPO = TMM + (10.0) LNG1272
10 TR = (T*1.8)/TCMP LNG1273
NUM = (2901.01) -((5738.92)*TR) +((2849.85)*TR*TR) LNG1274
1 + ((1.74127) / (1.01 - TR)) LNG1275
NUM = NUM * (TR - (1.0)) LNG1276
IF(NUM.LT.(-180.)) DD = 0 LNG1277
IF(NUM.LT.(-180.)) GO TO 11 LNG1278
DD = EXP(NUM) LNG1279
C THE STATEMENT ABOVE MAY RESULT IN AN UNDERFLOW SENSE LIGHT LNG1280
C ON SOME OPERATING SYSTEMS WHEN THE NUMBER NUM IS A LARGE LNG1281
C NEGATIVE NUMBER. THE LARGE NEGATIVE VALUE IS PROPER AND LNG1282
C THE CORRECT ANSWER FOR DD IS ZERO. LNG1283
11 TCMP = TMM + ((TMC-TMM)*DD) LNG1284
TEST = (1.0) - (TCMP/TPO) LNG1285
TEST = ABS(TEST) LNG1286
IF(TEST.LT.(.0001)) GO TO 12 LNG1287
ICT = ICT + 1 LNG1288
LNG1289
IF(ICT.GT.250) GO TO 12 LNG1290
TPO = TCMP LNG1291
GO TO 10 LNG1292
12 TMC = TCMP/(1.8) LNG1293
RETURN LNG1294
END LNG1295
SUBROUTINE BLOCK LNG1296
TYPE REAL KIJ,AJI LNG1297
DIMENSION KIJ(8,8),AJI(8,8) LNG1298
COMMON /PAR/ KIJ,AJI LNG1299
DATA KIJ/ LNG1300
*0.000000,.293E-07,.911E-06,.0030498,0.001000,0.001000,0.001000, LNG1301
*0.001000,1.088608,0.000000,.188E-10,.463E-06,.462E-08,.120E-03, LNG1302
*0.001000,0.001000,1.098880,1.078910,0.000000,.517E-06,.535E-06, LNG1303
*.188E-10,0.001000,0.001000,.8578232,1.146026,1.009559,0.000000, LNG1304
*.546E-06,.136E-05,0.001000,0.001000,0.995000,1.287762,.9868486, LNG1305
*.9690485,0.000000,.541E-06,0.001000,0.001000,1.178993,1.266327, LNG1306
*1.016231,.9976706,.9435714,0.000000,0.001000,0.001000,0.950000, LNG1307
*1.000000,1.000000,1.000000,1.000000,1.000000,0.000000,0.001000, LNG1308
*0.995000,1.000000,1.000000,1.000000,1.000000,1.000000,1.000000, LNG1309
*0.000000/ LNG1310
DATA AJI/ LNG1311
*0.000000,.161E-04,.257E-06,.0003462,0.001000,0.001000,0.001000, LNG1312
*0.001000,1.014967,0.000000,.188E-10,.197E-06,.184E-05,.335E-06, LNG1313
*0.001000,0.001000,1.008061,1.004105,0.000000,.142E-06,.103E-04, LNG1314
*.188E-10,0.001000,0.001000,.9863463,1.011399,1.001862,0.000000, LNG1315
*.105E-04,.119E-05,0.001000,0.001000,0.995000,1.024894,1.003289, LNG1316
*.9987221,0.000000,.119E-05,0.001000,0.001000,.9566020,1.025269, LNG1317
*1.005949,1.001225,.9968984,0.000000,0.001000,0.001000,0.995000, LNG1318
*1.000000,1.000000,1.000000,1.000000,1.000000,0.000000,0.001000, LNG1319
*0.995000,1.000000,1.000000,1.000000,1.000000,1.000000,1.000000, LNG1320
*0.000000/ LNG1321
END LNG1322

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American Gas Association, Inc. (see page 1).

10. SUPPLEMENTARY NOTES

Document describes a computer program; SF-185, FIPS Software Summary, is attached.

11. ABSTRACT (A 200-word or less factual summary of most significant information. If document includes a significant bibliography or literature survey, mention it here)

Four mathematical models of the equation of state for LNG like mixtures are presented. The four models include an extended corresponding states model, a cell model, a hard sphere model and a revised Klosek and McKinley model. Each of the models has been optimized to the same experimental data set which included data for pure nitrogen, methane, ethane, propane, iso and normal butane, iso and normal pentane and mixtures thereof. For LNG like mixtures (mixtures of the orthobaric liquid state at temperatures of 120 K or less and containing at least 60% methane, less than 4% nitrogen, less than 4% each of iso and normal butane and less than 2% total of iso and normal pentane), all of the models are estimated to predict densities to within 0.1% of the true value. The revised Klosek and McKinley model is valid only for mixtures within the range of temperature and composition specified above while the other three models are valid for a broader range of pressure, temperature and composition. The experimental PVTx data set used in the optimization together with comparisons are given and listings of computer programs for each of the models are included.

2. KEY WORDS (Six to twelve entries; alphabetical order; capitalize only proper names; and separate key words by semicolons)

Cell model; comparisons; computer programs; corresponding states; equation of state; hard sphere; LNG; mixtures; PVTx data; revised Kosek and McKinley.

3. AVAILABILITY

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